Stoichiometry Worksheet # 1

 Glucose is used as a source of energy by the human body. The overall reaction in the body is C6H12O6+6O2 → 6CO2 + 6H2O

Calculate the number of moles of oxygen needed to oxidize 12.5 g of C6H12O6.

Ammonia is synthesized from hydrogen and nitrogen according to the following equation.

$$N_2 + 3H_2 \rightarrow 2NH_3$$

If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH₃) can be produced?

3) Assume that in the decomposition of potassium chlorate, KCLO₃,

80.5 moles of O2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,

9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" PtCl₂(NH₂)₂, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction

$$K_2PtCl_4 + 2NH_3 \longrightarrow 2KCl + PtCl_2(NH_3)_2$$

- A. How many moles of "cisplatin" can be produced from 2.50 g K2PtCl4?
- B. How many moles of NH3 would be needed if 3.8 moles of K2PtCl4 were present?

Chapter Stoichiometry Work Answers

Arnold B. Loebel

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web may 6 2019 drawing from susan sontag s seminal 1964 essay notes on camp this multifaceted publication presents the sartorial manifestations of the camp sensibility while contributing new

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web may 6 2019 this year s theme is camp notes on fashion curators in charge andrew bolton and wendy yu of the costume institute were inspired by susan sontag s 1964 essay notes on camp and the exhibit examines how the elements of irony humour parody pastiche artifice theatricality and exaggeration are expressed in fashion

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web may 7 2019 the first english definition of the term which appeared in a 1909 edition of the oxford english dictionary conformed to popular contemporary notions of camp ostentatious exaggerated

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