

Stuff I Should Know (Page 2)

Complex Ions & Common Ligands

Ligands	polar molecules & anions	NH_3 , H_2O , OH^- , CN^- , Cl^-	Odd example: $\text{Fe}^{3+} + \text{SCN}^- \rightleftharpoons \text{FeSCN}^{2+}$
Central ions	transition metals and Al^{3+}	Ag^+ , Cu^{2+} , Ni^{2+} , Zn^{2+} , etc. & Al^{3+}	
Examples	usually twice the number of ligands as the charge on the central ion. Key Words: "excess, concentrated"	$\text{Ag}(\text{CN})_2^-$, $\text{Cu}(\text{NH}_3)_4^{2+}$, $\text{Ni}(\text{OH})_4^{2-}$, $\text{Zn}(\text{NH}_3)_4^{2+}$, $\text{Al}(\text{OH})_4^-$	Reaction with Acid: $\text{Cu}(\text{NH}_3)_4^{2+} + \text{H}^+ \rightarrow \text{Cu}^{2+} + \text{NH}_4^+$

Organic Chemistry & Functional Groups

alkanes $\text{C}_n\text{H}_{2n+2}$	alkenes C_nH_{2n}	alkynes $\text{C}_n\text{H}_{2n-2}$	aromatics (benzene) C_6H_6
alcohol $\text{R}-\text{OH}$	aldehyde $\text{R}-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{H}$	ketone $\text{R}-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{R}$	ether $\text{R}-\text{O}-\text{R}$
carboxylic acid $\text{R}-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{OH}$	ester $\text{R}-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{O}-\text{R}$	amine $\text{R}-\text{NH}_2$	amide $\text{R}-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{NH}_2$
Substituted benzene:	ortho = 1,2	meta = 1,3	para = 1,4

nuclear chem

alpha ${}^4_2\text{He}$
beta electron ${}^0_{-1}\text{e}$
neutron ${}^1_0\text{n}$
positron ${}^0_{+1}\text{e}$

ΔH ΔS Spont?

-	+	at all temps
+	+	high temps
-	-	low temps
+	-	no temps

Note: ΔS in J
 ΔG & ΔH in kJ

K_{sp} & Solubility, s

1:1	$K_{sp} = s^2$
1:2	$K_{sp} = 4s^3$
1:3	$K_{sp} = 27s^4$
2:3	$K_{sp} = 108s^5$

Lewis Acids & Bases



acid anhydrides (oxides of nonmetals, CO_2)

basic anhydrides (oxides of metals, MgO)



decomposition reactions: $\text{MgCO}_3 \rightarrow \text{MgO} + \text{CO}_2$

Strange Examples: $\text{P}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{PO}_4$

Strange Ions: (nitride, N^{3-}) (hydride, H^-)



Flame Test Colors

Barium - green
Sodium - yellow
Copper - blue (w/ green)
Potassium - lavender
Strontium - red
Lithium - red
Calcium - orange

Quantum Numbers

n	1, 2, 3, ...
l	0 ... (n-1)
m	-l ... +l
m_l	+1, -1
l	0 = s, 1 = p, 2 = d, 3 = f

Writing Lewis Structures

hint: use one valence electron to connect F's or Cl's then determine lone pairs (Ex: XeF_4)

Product-Favored (Spontaneous) Reactions

$$\Delta G < 0 \quad E^\circ > 0 \quad K_{eq} > 1$$

Properties Indicate Strength of Intermolecular Forces (IMF's)

IMF	BP	FP	H_{vap}	H_{fus}	VP
IMF	BP	FP	H_{vap}	H_{fus}	VP

Orders of Reactions & Graphs That Give Straight Lines

0 th Order	1 st Order	2 nd Order
$[\text{R}]$ vs. Time	$\ln[\text{R}]$ vs. Time	$1/[\text{R}]$ vs. Time
slope = -k	slope = -k	slope = -k

Electrochemical Cells

anode	cathode
oxidation	reduction
- side	+ side
lower E°	higher E°
e^- leave	e^- enter

Bond Orders

bond	B.O.	σ
single	1	σ
double	2	$\sigma + \pi$
triple	3	$\sigma + 2\pi$

SN & hybridization & shape

Steric Number	hybridization	basic shape
1	s	—
2	sp	linear
3	sp ²	Δ planar
4	sp ³	tetrahedral
5	sp ³ d	Δ bipyramidal
6	sp ³ d ²	octahedral

IMF's

London	nonpolar molecules, ex: CH_4 , He
dipole-dipole	polar molecules, ex: H_2S , SO_2
hydrogen bonding	H-F, H-O-, H-N-, NH_3 , H_2O amines and alcohols
metallic	metals, Ag, Pb
ionic	salts, NaCl , CaCO_3 (Note: "ates" contain covalent bonds)
covalent network	C (graphite), C (diamond), SiO_2 , WC, Si, SiC (Note: graphite = London, too)

Activity of Metals (Four Groups)

Metals	React with...
Groups I & II	H_2O ex: $\text{Li} + \text{H}_2\text{O} \rightarrow \text{Li}^+ + \text{OH}^- + \text{H}_2$
all others	Non-oxidizing Acid, ex: HCl $\text{Zn} + 2\text{HCl} \rightarrow \text{H}_2 + \text{ZnCl}_2$
Cu, Ag, Hg	Oxidizing Acid, HNO_3 or H_2SO_4 (conc.) $\text{Cu} + \text{HNO}_3 \rightarrow \text{NO}_2 + \text{H}_2\text{O} + \text{Cu}^{2+}$
Au, Pt, Ir	Aqua Regia (HNO_3 + HCl)

Cheat Sheet For Chemistry Final

Briton Hadden, Henry Robinson Luce



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Cracking the SAT Chemistry Subject Test, 2013-2014 Edition Princeton Review, Theodore Silver, M.D., 2013-03-05 Offers test strategies reviews key concepts of chemistry and provides three full length practice tests with answers and explanations

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The Blight on the Ivy Richard E. Gordon, Katherine K. Gordon, 1963 Fml Shaun David Hutchinson, 2013-06-25 At a party near the end of senior year seventeen year old Simon Cross imagines his life with and without Cassie the girl he has yearned for since they were freshman and begins to discover the unpredictable wonders of life his best friends Ben and Coop have urged him to explore Kennedy Stern Christian Suspense Complete Box Set (Books 1-9) Alana Terry, 2018-06-28 College is easy for missionary kid Kennedy Stern It s staying alive that s a lot more difficult Kennedy could thrive in Harvard s pre med program if only there weren t so many people who wanted her dead Engrossing thought provoking and never preachy the Kennedy Stern Christian suspense series follows Kennedy as an undergrad who gets kidnapped while volunteering at a crisis pregnancy center experiences police brutality firsthand finds herself on a hijacked airplane and much more Alana Terry is one of the few authors that doesn t create a bad book Her stories are crafted with a lot of insight and tackle issues most authors wouldn t dream of touching Sheila McIntyre book reviewer Find out why the Kennedy Stern novels are being called the most relevant Christian series of the decade Buy the complete nine book library today but be prepared to stay up late **Time** Briton Hadden, Henry Robinson Luce, 1924 Reels for 1973 include Time index 1973

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Cases argued and determined in the courts of Iowa, Michigan, Minnesota, Nebraska, North Dakota, South
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