

Science

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(Chapter – 1) (Chemical Reactions and Equations) (Intext)
(Class 10)

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Question 1:

A solution of a substance 'X' is used for white washing.

(i) Name the substance 'X' and write its formula.

(ii) Write the reaction of the substance 'X' named in (i) above with water.

Answer 1:

(i) The substance 'X' is calcium oxide. Its chemical formula is CaO.

(ii) Calcium oxide reacts vigorously with water to form calcium hydroxide (slaked lime)



Question 2:

Why is the amount of gas collected in one of the test tubes in Activity 1.7 double of the amount collected in the other? Name this gas.

Answer 2:

During the Electrolysis of water, hydrogen and oxygen is get separated by the electricity. Water (H₂O) contains two parts hydrogen and one part oxygen. Since hydrogen goes to one test tube and oxygen goes to another, the amount of gas collected in one of the test tubes is double of the amount collected in the other.

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Question 1:

Why does the colour of copper sulphate solution change when an iron nail is dipped in it?

Answer 1:

When an iron nail is dipped in a copper sulphate solution, iron (which is more reactive than copper) displaces copper from copper sulphate solution forming iron sulphate, which is green in colour.



Therefore, the blue colour of copper sulphate solution fades and green colour appears.

Question 2:

Give an example of a double displacement reaction other than the one given in Activity 1.10.

Answer 2:

Sodium carbonate reacts with calcium chloride to form calcium carbonate and sodium chloride.



In this reaction, sodium carbonate and calcium chloride exchange ions to form two new compounds. Hence, it is a double displacement reaction.

Question 3:

Identify the substances that are oxidised and the substances that are reduced in the following reactions.

(i) $4\text{Na(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{Na}_2\text{O(s)}$

(ii) $\text{CuO(s)} + \text{H}_2\text{(g)} \rightarrow \text{Cu(s)} + \text{H}_2\text{O(l)}$

Answer 3:

(i) Sodium (Na) is oxidised as it gains oxygen and oxygen gets reduced.

(ii) Copper oxide (CuO) is reduced to copper (Cu) while hydrogen (H₂) gets oxidised to water (H₂O).

Chapter Section reactions And Equations Answers

Arnold B. Loebel



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fw php a collection of php utilities kandi - Mar 30 2022

web fw php is a php library fw php has no bugs it has no vulnerabilities and it has low support you can download it from github read the documentation in the docs directory for more information support quality security license reuse support fw php has a low active ecosystem it has 1 star s with 0 fork s

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web php dosya İşlemleri fopen ile bir dosya açmak bu fonksiyon bir dosyanın içeriğine erişmek dosya üzerinde işlem yapmak veri eklemek için kullanılır kullanım Şekli fopen dosya özelliklik

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web fw php can be configured in 2 ways first one is using only php and second one is loading configuration from json file in examples below we will load cookie php extension from plugins folder change directory of public files from default public to new public dir set route to index function using php only this is basic configuration from

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web here is my code to write the file filename mysql fetch array query pagename fw fopen filename w true fwrite php id id include admin renderpage php fclose fw

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web php yüklemek için buraya tıklayın evde bilgisayarınızda localhost yerel sunucu kurmak ve php yi çalışır hale getirmek için tüm kurulumları gerçekleştirin ayrıca xampp ve easyphp kurarak da kolaylıkla bahsedeceğimiz kodları çalıştırabilir ve denemeler yapabilirsiniz uyari windows 7 de php çalıştırmak için zip