

Dissolves noble metals



98. What happens when hydrochloric acid is treated with NH_3 ?

1M

Answer: $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$ (White fumes)

99. Write the structure of

i) Hypochlorous acid ii) Chlorous acid iii) Chloric acid

iv) Perchloric acid.

3M each

Answer:



Answer:

100. What are interhalogen compounds? Give an example. Why they are more reactive than individual halogens?

3M

Answer: When two different halogen atoms react inter halogen compounds are formed. Eg: ClF_3 , ICl , BrF_3 , IF_5 . Reactivity is more compared with halogens because X-X' bond is weaker than X-X bond in pure halogens.

101. How is following interhalogen compound prepared?

i) ClF_3 ii) ICl_3 iii) BrF_3

1M each

Answer:



(excess)

102. Name i) the radioactive noble gas ii) most abundant noble gas.

1M

Answer: i) Radon ii) Argon

103. Why noble gases are chemically inert?

1M

Answer: Stable completely filled orbitals are there.

104. Why noble gases have maximum ionization enthalpy in the corresponding period.

1M

Answer: Stable completely filled orbitals are there

105. Why noble gases have positive electron gain enthalpy

1M

Answer: Stable completely filled orbitals are there

106. Which is the first noble gas compound synthesized?

1M

Answer: $\text{Xe}^+\text{PtF}_6^-$

107. Who prepared first noble gas compound?

1M

Answer: Neil Bartlett

108. Write the chemical equations to prepare following compounds with the conditions required.

i) XeF_4 ii) XeO_3 iii) XeO_2F_2

1M each



109. Write/ Name the structure of

i) XeF_2 ii) XeF_4 iii) XeF_6 iv) XeOF_4 v) XeO_3

1M each

Answer: i) linear ii) square planar iii) Distorted octa hedral

iv) Square pyramidal) trigonal pyramidal

110. Noble gases have very low boiling point. Why?

1M

Answer: They are mono atomic due to weak dispersion forces, hence have low boiling points.

Chapter Wise Important Questions In Chemistry

Priti Singhal

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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