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CHAPTER 9 REVIEW

Stoichiometry

MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided.

Given the following equation: C₃H₄(g) + xO₂(g) → 3CO₂(g) + 2H₂O(g)

a. What is the value of the coefficient x in this equation?

b. What is the molar mass of C3H4?

c. What is the mole ratio of O₂ to H₂O in the above equation?

d. How many moles are in an 8.0 g sample of C₃H₄?

e. If z mol of C₃H₄ react, how many moles of CO₂ are produced, in terms of z?

2. a. What is meant by ideal conditions relative to stoichiometric calculations?

b. What function do ideal stoichiometric calculations serve?

c. Are actual yields typically larger or smaller than theoretical yields?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

Assume the reaction represented by the following equation goes all the way to completion:

 $N_2 + 3H_2 \rightarrow 2NH_3$

a. If 6 mol of H₂ are consumed, how many moles of NH3 are produced?

b. How many grams are in a sample of NH₃ that contains 3.0 × 10²³ molecules?

Chapter Review Stoichiometry Mixed

G Thomas

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Chemically Deposited Nanocrystalline Metal Oxide Thin Films Fabian I. Ezema, Chandrakant D. Lokhande, Rajan Jose, 2021-06-26 This book guides beginners in the areas of thin film preparation characterization and device making while providing insight into these areas for experts As chemically deposited metal oxides are currently gaining attention in development of devices such as solar cells supercapacitors batteries sensors etc the book illustrates how the chemical deposition route is emerging as a relatively inexpensive simple and convenient solution for large area deposition. The advancement in the nanostructured materials for the development of devices is fully discussed Perovskites and Related Mixed Oxides Pascal Granger, Vasile I. Parvulescu, Serge Kaliaguine, Wilfrid Prellier, 2016-02-23 This comprehensive handbook and ready reference details all the main achievements in the field of perovskite based and related mixed oxide materials. The authors discuss in an unbiased manner the potentials as well as the challenges related to their use thus offering new perspectives for research and development on both an academic and industrial level The first volume begins by summarizing the different synthesis routes from molten salts at high temperatures to colloidal crystal template methods before going on to focus on the physical properties of the resulting materials and their related applications in the fields of electronics energy harvesting and storage as well as electromechanics and superconductivity The second volume is dedicated to the catalytic applications of perovskites and related mixed oxides including but not limited to total oxidation of hydrocarbons dry reforming of methane and denitrogenation The concluding section deals with the development of chemical reactors and novel perovskite based applications such as fuel cells and high performance ceramic membranes Throughout the contributions clearly point out the intimate links between structure properties and applications of these materials making this an invaluable tool for materials scientists and for catalytic and physical chemists Introductory Chemistry Darrell D. Ebbing, Albert E. Richardson, 1995

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