Chapter 3 - Stoichiometry

Section 3.1: Counting by Weighing

Section 3.2 – Atomic mass

<u>Mass spectrometer</u>: the most accurate method available for comparing the masses of atoms.

- Atoms or molecules are passed into a beam of high-speed electron, so they changed into positive
 ions, which accelerated by an electric field into a magnetic field changes the path of the ions, the
 amount of that path deflection depends on the mass of the ion (more mass, less deflection)
- A comparison of the position where the ions hit the detector plate gives very accurate values of their relative masses. (¹²C and ¹³C are analyzed in mass spectrometer ratio Mass ¹³C/ Mass ¹²C=1.0836129)
- There is no C atoms with atomic weight = 12.011 (because this is the average of the masses of the carbon isotopes.

Relative atomic mass: is the weighted average of one atom f an element relative to one-twelfth of an tom of carbon-12.

Average atomic mass = % x atomic mass of first isotope + % x mass of second isotope +

- <u>Problem 1</u>: A hypothetical element consists of two isotopes of masses 69.95 amu and 71.95 amu with abundances of 25.7% and 74.3%, respectively. What is the average atomic mass of this element? (71.4)
- <u>Problem 2</u>: The atomic mass of rhenium is 186.2. Given that 37.1% of natural rhenium is rhenium-185, what is the other stable isotope? (187)

Section 3.3 - The mole

- Molar Mass the mass in grams of one mole of the compound. (also called molecular weight)
- Mole the number equal to the number of carbon atoms in exactly 12 grams of pure 12C.
- Avogadro's number 6.022 x 10²³

Problem1: What is the mass of 4 atom(s) of copper in grams?

Section 3.4 - molar mass

No of moles
$$=\frac{Mass}{Molar\ mass} = \frac{no.of atoms, ions, or\ molecules}{6.022 \times 1023} = \frac{volume}{22.4}$$

Section 3.5 – learning to solve problems

Section 3.6 – Percent composition of compounds:

$$Mass percent = \frac{mass}{total \ mass} \times 100$$

The sum of individual mass percent values = 100%

Chapter Review Stoichiometry Section 3

RM Cervero

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Holt Chemistry Ralph Thomas Myers, 2004 MCAT Comprehensive Review, 2004 Kaplan, Inc, 2003 The bestselling MCAT prep guide is a must for pre med students preparing for the demanding Medical College Admissions Test MCAT The CD features one full length practice test with detailed explanations plus effective tips and strategies Films Hari Singh Nalwa, 2001-11-17 This five volume handbook focuses on processing techniques characterization methods and physical properties of thin films thin layers of insulating conducting or semiconductor material. The editor has composed five separate thematic volumes on thin films of metals semimetals glasses ceramics alloys organics diamonds graphites porous materials noncrystalline solids supramolecules polymers copolymers biopolymers composites blends activated carbons intermetallics chalcogenides dyes pigments nanostructured materials biomaterials inorganic polymer composites organoceramics metallocenes disordered systems liquid crystals quasicrystals and layered structures Thin films is a field of the utmost importance in today s materials science electrical engineering and applied solid state physics with both research and industrial applications in microelectronics computer manufacturing and physical devices Advanced high performance computers high definition TV digital camcorders sensitive broadband imaging systems flat panel displays robotic systems and medical electronics and diagnostics are but a few examples of miniaturized device technologies that depend the utilization of thin film materials The Handbook of Thin Films Materials is a comprehensive reference focusing on processing techniques characterization methods and physical properties of these thin film materials MCAT Comprehensive Review 1999 ,1998-11 Complete Review of ALL the science you need to know Full Length Practice Test with complete explanations for each question 2 Full Length Verbal Reasoning Tests with detailed explanations and strategies **Complete Preparation** for the MCAT Williams & Wilkins Review, 1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam's content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT MCAT Comprehensive A Complete Preparation for the MCAT Aftab S. Hassan, James L. Flowers, 1992 **Review** ,2004 **Chemistry** Oxtoby,1994 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress

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Table of Contents Chapter Review Stoichiometry Section 3

- 1. Understanding the eBook Chapter Review Stoichiometry Section 3
 - The Rise of Digital Reading Chapter Review Stoichiometry Section 3
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter Review Stoichiometry Section 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter Review Stoichiometry Section 3
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter Review Stoichiometry Section 3
 - Personalized Recommendations
 - $\circ\,$ Chapter Review Stoichiometry Section 3 User Reviews and Ratings
 - $\circ\,$ Chapter Review Stoichiometry Section 3 and Bestseller Lists
- 5. Accessing Chapter Review Stoichiometry Section 3 Free and Paid eBooks
 - \circ Chapter Review Stoichiometry Section 3 Public Domain eBooks
 - Chapter Review Stoichiometry Section 3 eBook Subscription Services
 - Chapter Review Stoichiometry Section 3 Budget-Friendly Options

- 6. Navigating Chapter Review Stoichiometry Section 3 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Chapter Review Stoichiometry Section 3 Compatibility with Devices
 - Chapter Review Stoichiometry Section 3 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Chapter Review Stoichiometry Section 3
 - Highlighting and Note-Taking Chapter Review Stoichiometry Section 3
 - Interactive Elements Chapter Review Stoichiometry Section 3
- 8. Staying Engaged with Chapter Review Stoichiometry Section 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter Review Stoichiometry Section 3
- 9. Balancing eBooks and Physical Books Chapter Review Stoichiometry Section 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter Review Stoichiometry Section 3
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter Review Stoichiometry Section 3
 - Setting Reading Goals Chapter Review Stoichiometry Section 3
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter Review Stoichiometry Section 3
 - Fact-Checking eBook Content of Chapter Review Stoichiometry Section 3
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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