

## Chapters 1-2

- Mass
- Weight
- Independent/Dependent Variable
- Theory vs. Law
- Lab Safety
- SI/Metric System → Be able to convert on the spot on your own
- Derived Unit
- Dimensional Analysis → You will be given the unit conversions
- Temperature Calculations
- Scientific Notation
- Accuracy vs. Precision
- Significant Figures → Know the rules!
  - Addition/Subtraction
  - Multiplication/Division
- Be able to create a graph with the appropriate labels
  - Be able to calculate slope

## Chapter 3

- States of matter
- Physical/chemical change
- Heterogeneous/homogeneous mixtures
- Separation techniques
- Percent mass
- Elements – name/symbol

## Chapter 4

- Dalton's atomic theory
- Know the gentlemen that led to the discovery of the subatomic particles. You will be asked for their name, experiment (except for the neutron), and what subatomic particle they discovered
- Atom, proton, neutron, electron, nucleus
- Be able to calculate
  - Average atomic mass
  - Mass number
  - Number of neutrons
  - Number of electrons
    - With the +/- charges & what that means; with the number of electrons
- Use your periodic table to find information

## Chapter 5

- Electromagnetic spectrum

# Chemistry Study Guide Midterm Fall 2013

**Jill Kirsten Robinson, John E.  
McMurry, Robert C. Fay**



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**Implementation and Critical Assessment of the Flipped Classroom Experience** Scheg, Abigail G., 2015-01-31 In the past decade traditional classroom teaching models have been transformed in order to better promote active learning and learner engagement Implementation and Critical Assessment of the Flipped Classroom Experience seeks to capture the momentum of non traditional teaching methods and provide a necessary resource for individuals who are interested in taking advantage of this pedagogical endeavor Using narrative explanations and foundation materials provided by experienced instructors this premier reference work presents the benefits and challenges of flipped methodology implementation in today's classroom to educators and educational administrators across all disciplines and levels

**Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

*Introductory Chemistry* Mike Lucarelli, David Alan Proctor, 2002

**Chemistry and Student Study Guide and Study Guide, Fifth Edition and Smarthinking and Letter** Zumdahl, 2002-01-01

*Chemistry for Changing Times* John W. Hill, 1988

*Chemical Principles Student's Study Guide & Solutions Manual* John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

**Study Guide for Chemistry H** LeMay, Theodore Brown, Bruce Bursten, Catherine Murphy, Patrick Woodward, Matthew Stoltzfus, 2017-03-21 This guide assists students through the text material with chapter overviews learning objectives a review of key terms as well as self tests with answers and explanations This student guide also features MCAT practice questions

**Study Guide for Chemistry** Jill Kirsten Robinson, John E. McMurry, Robert C. Fay, 2019-01-04

**Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1992

*Student Study Guide for Chemistry* Kenneth Goldsby, Professor, Raymond Chang, Dr., 2015-03-06 This valuable ancillary contains material to help the student practice problem solving skills For each section of a chapter the author provides study objectives and a summary of

the corresponding text Following the summary are sample problems with detailed solutions Each chapter has true false questions and a self test with all answers provided at the end of the chapter

**Chemistry, Study Guide** Neil D. Jespersen, James E. Brady, Alison Hyslop, 2011-05-03 Chemistry The Study of Matter and Its Changes Sixth Edition will provide the necessary practice support and individualised instruction that ensures success in the General Chemistry course This text provides the forum for problem solving and concept mastery of chemical phenomena that leads to proficiency and success in the General Chemistry course This edition will continue a molecular basis of chemistry tradition but in a manner that overtly and repeatedly reinforces the way properties at the molecular level are related to properties we observe at the macroscopic level The unique chemical tools approach employed in this book provides a way of thinking that helps those students develop the ability to analyse and solve both mathematical and conceptual problems This text follows the successful three step approach described as ANALYSIS SOLUTION and IS THE ANSWER REASONABLE This encourages the student to think about the problem before attempting to solve it then working through the solution and finally asking the important question Does the answer make sense There are problem sets called Bringing It Together that contain problems which require students to bring together concepts from two or more of the preceding chapters This reinforces learned concepts and builds concept mastery

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