

AP CHEM Unit 4

UNIT 4 HOMEWORK KEY

Unit 4: Chemical Equilibrium

1. Identify the two substances in each equilibrium by state of matter. *Changes:*

$\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 Fe: solid, Fe₃O₄: solid, H₂O: gas, H₂: gas

2. Predict the direction of change in the concentration of each species if the following changes occur. *Changes:*

a. Addition of a reactant: $\text{H}_2\text{O}(\text{g})$ increases
 b. Removal of a product: $\text{H}_2(\text{g})$ decreases
 c. Addition of a catalyst: no change
 d. Increase in volume: all concentrations decrease

3. Write and balance the following equations.

a. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 b. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 c. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 d. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 e. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$
 f. $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{Fe}_3\text{O}_4(\text{s}) + \text{H}_2(\text{g})$

4. Predict the products of the following reactions. You don't need to add coefficients.

a. $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$
 b. $\text{Ca}(\text{OH})_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
 c. $\text{Fe} + \text{Cl}_2 \rightarrow \text{FeCl}_3$
 d. $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$

5. How can you identify a **SALT** by looking at its formula?

6. How can you identify a **MOLECULE** by looking at its formula?

Answer Keys are DETAILED!

Fully Editable!

Chemistry Unit 4 Question Packet

L Manion

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