# Unit 5 - Counting Particles - Objectives

| Re | view Concepts  |  |
|----|--|--|
|    | <ul> <li>Types of substances</li> <li>Chemical formulas of<br/>substances (U 4)</li> </ul>   |  |
| 1. | State evidence for Avogadro's<br>Hypothesis. Use Avogadro's<br>Hypothesis and experimental<br>data to determine the relative<br>mass of molecules. |  |
| 2. | Use experimental data to<br>determine the relative mass<br>of two objects.   |  |
| 3. | Use experimental data to<br>determine the number of<br>items in a sample without<br>actually counting them.  |  |
| 4. | Given the chemical formula of<br>a substance, determine the<br>molar mass.   |  |
| 5. | Given the mass of a<br>substance, determine<br>a. the number of moles of the<br>sample<br>b. the number of atoms or<br>molecules in the sample     |  |

# **Chemistry Unit 5 Counting Particles Key**

Steven S. Zumdahl

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