

## CP Chemistry – Unit 3 Review

To prepare to do well on the Unit 3 test, you should assemble your notes, the objectives, worksheets, labs and activity notes, and the quizzes and review them, preferably in a small group where you can draw from each other's understanding. Here are the key points you should know.

### Energy

Think of energy as a quantity that is always involved when there is a change in the state of matter. When a substance gets hotter or colder or changes phase, energy is either transferred into or out of the system. The two key ways energy is stored is **kinetic** (due to the motion of the particles) and **interaction** (due to attractions between the particles). Remember that attractions lower the energy state, so one must add energy to a system to pull particles apart. The three ways that energy is transferred is by heating (Q), working (W) and radiating (R); this course focuses on Q. You will be expected to be able to:

- I. Draw energy bar graphs to account for energy storage and transfer in all sorts of changes. Make up a sample situation and sketch the bar graph. (review ws 1 and 2, quiz)



# Chemistry Unit 3 Review Answer Key

**Guillaume Favre**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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