

AP Chemistry — Unit 6 Progress Check: MCQ Exam Questions and Answers Updated (2024/2025) (Verified Answers)

1.

The chemical reaction between HCl(aq) and $\text{NH}_3(\text{aq})$ is represented above. A student combines equimolar amounts of HCl(aq) and $\text{NH}_3(\text{aq})$, both solutions initially at 24°C , in a coffee-cup calorimeter. The student observes that the mixture reaches a temperature of 28°C . Based on the experimental results, which of the following can be concluded about the reaction? -

ANSWER C. It is an exothermic process, because energy is released by the reaction and is gained by the reaction mixture.

2.

When pellets of NaOH(s) are added to a flask of water, it is observed that the temperature of the water increases as the pellets dissolve. Which of the following claims about the observed dissolution of NaOH(s) in water is most accurate? - ANSWER C. It is an exothermic process because heat energy is absorbed by the water as the NaOH(s) dissolves in it.

3.

An advertisement for a commercial hand warmer claims that the hand warmer works because a chemical reaction in the hand warmer draws out the body's own natural heat, causing a warming effect. Which of the following states the accuracy of the claim in the advertisement and best provides a correct scientific justification of the claim? - ANSWER A. The advertisement's claim is inaccurate because heat flowing from the hands to the warmer would only happen if the chemical reaction was endothermic, which would cause the hands to feel colder.

4.

A student carefully drops a 9.0g solid Zn pellet initially at 50.0°C into an insulated cup containing 30.0g of water at 27.8°C . The student predicts that the temperature of the water will increase after the pellet is added. Which of the following statements is the best justification for the student's prediction? - ANSWER B. Collisions between the water molecules and the surface of the Zn pellet will result in the transfer of energy, increasing the average kinetic energy of the water molecules.

5.

Two aqueous solutions of equal volume and concentration were kept in flasks and held at different temperatures. The two solutions were combined in a larger flask. Based on this information, which of the following predictions is correct? - ANSWER B. The average kinetic energy of the particles in the cooler solution will increase as they collide with the particles from the warmer solution.

6.

Chemistry Unit 6 Answers

Roger Norris



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web au cours de ces dernières années le danemark est devenu l un des pays les plus riches d europe l économie du danemark est très dépendante du commerce extérieur l une des activités majoritaire du pays est l agriculture avec de vastes étendues de terres agricoles et plus de 50 000 agriculteurs le danemark est l un des plus

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web mar 16 2020 un modèle économique et social fragile mais à suivre le danemark est un pays d europe du nord et le plus petit des pays scandinaves il est membre de l union européenne depuis 1973 avec pour capitale copenhagen il partage des frontières avec l allemagne au sud et la suède à l est par le pont de l Øresund qui relie