

## SNC2D Chemistry Unit Test Practice

### Multiple Choice (1 mark each):

- Which of the following will form positive ions?  
(A) the alkali metals    (B) the halogens    (C) the noble gases    (D) all of the above
- Which of the following compounds is an ionic compound?  
(A)  $\text{CO}_2$     (B)  $\text{CaO}$     (C)  $\text{CH}_4$     (D) all of the above
- Which of the following compounds is a molecular compound?  
(A)  $\text{PCl}_3$     (B)  $\text{PbCl}_2$     (B)  $\text{Pb}(\text{ClO}_3)_2$     (D) all of the above
- The Roman numeral in the name iron (III) nitride indicates:  
(A) the number of iron atoms in the compound  
(B) the number of nitrogen atoms in the compound  
(C) the valence of the iron  
(D) the valence of the nitrogen
- The bond in a molecule of  $\text{Cl}_2$  could be described as:  
(A) covalent    (B) ionic    (C) both (A) and (B)    (D) neither (A) nor (B)
- If a strip of magnesium reacts with oxygen gas, the mass of the resulting magnesium oxide will be \_\_\_\_\_ the mass of the magnesium.  
(A) less than    (B) more than  
(C) the same as    (D) It cannot be determined.
- In an equation representing a chemical reaction, the reactants are always written:  
(A) to the left of the arrow    (B) to the right of the arrow  
(C) above the arrow    (D) below the arrow
- Which of the following is an indication a chemical reaction has occurred?  
(A) the formation of a precipitate    (B) a colour change  
(C) a temperature change    (D) all of the above
- If  $\text{AlBr}_3$  reacts with  $\text{Cl}_2$ , the products of the reaction will be:  
(A)  $\text{AlCl}_2$  and  $\text{Br}_2$     (B)  $\text{AlCl}_3$  and  $\text{Br}_2$     (C)  $\text{Al}$  and  $\text{Br}_2\text{Cl}_2$     (D)  $\text{Al}$  and  $\text{BrCl}$
- The reaction in Question 9 could be described as a \_\_\_\_\_ reaction.  
(A) decomposition    (B) double displacement  
(C) single displacement    (D) synthesis

# Chemistry Unit Test Practice

**Hongru Du**

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## **Table of Contents Chemistry Unit Test Practice**

1. Understanding the eBook Chemistry Unit Test Practice
  - The Rise of Digital Reading Chemistry Unit Test Practice
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit Test Practice
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Unit Test Practice
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit Test Practice
  - Personalized Recommendations
  - Chemistry Unit Test Practice User Reviews and Ratings
  - Chemistry Unit Test Practice and Bestseller Lists
5. Accessing Chemistry Unit Test Practice Free and Paid eBooks
  - Chemistry Unit Test Practice Public Domain eBooks

- Chemistry Unit Test Practice eBook Subscription Services
- Chemistry Unit Test Practice Budget-Friendly Options
- 6. Navigating Chemistry Unit Test Practice eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Unit Test Practice Compatibility with Devices
  - Chemistry Unit Test Practice Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Unit Test Practice
  - Highlighting and Note-Taking Chemistry Unit Test Practice
  - Interactive Elements Chemistry Unit Test Practice
- 8. Staying Engaged with Chemistry Unit Test Practice
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Unit Test Practice
- 9. Balancing eBooks and Physical Books Chemistry Unit Test Practice
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Unit Test Practice
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Unit Test Practice
  - Setting Reading Goals Chemistry Unit Test Practice
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Unit Test Practice
  - Fact-Checking eBook Content of Chemistry Unit Test Practice
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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