Introduction to Balancing Equations

Draw a line down the middle of your book and add the title of "Reactants" to the left hand side and "Products" to the right hand side.

For each of the following equations you need to:

- Draw the molecule for both Reactants and Products on the correct side of the paper.
- Make multiple of the molecules you will need to balance the number of atoms on each side of the reaction.
- Use coefficients in the given equations to describe the balanced reaction you have made.

(a). C +
$$O_2 \rightarrow CO_2$$

(b)
$$N_2 + O_2 \rightarrow NO$$

(c)
$$N_2 + H_2 \rightarrow NH_3$$

(d) C +
$$I_2 \rightarrow CI_4$$

(e)
$$CH_4 + O_2 \rightarrow CO_2 + H_2O$$

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Michael Hoelscher

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