

ChemQuest 1.1

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolyte Review

1. Write the formula for the following compounds:

- a) barium phosphate: $\text{Ba}_3(\text{PO}_4)_2$ b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $\text{Mg}_3(\text{PO}_4)_2$ d) lithium nitrate: LiNO_3

2. Write the name for the following compounds:

- a) $\text{Cu}(\text{NO}_3)_2$: copper(II) nitrate b) FeCl_3 : iron(III) chloride
 c) SnSO_4 : tin(II) sulfate d) SnF_4 : tin(IV) fluoride

Chemical Reactions: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that you can write to describe what happens above:

Chemical reactions result with atoms and molecules and electron and energy.

There are also some characteristics for chemical reactions and the reaction conditions. Here are a few points:

Critical Thinking Questions

1. Write the following reaction, with the reactants and products:



Chemical reactions result with atoms and molecules and electron and energy.

Chemquest 10 Answer Key

Christian G. Meyer



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This book delves into Chemquest 10 Answer Key. Chemquest 10 Answer Key is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemquest 10 Answer Key, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chemquest 10 Answer Key
 - Chapter 2: Essential Elements of Chemquest 10 Answer Key
 - Chapter 3: Chemquest 10 Answer Key in Everyday Life
 - Chapter 4: Chemquest 10 Answer Key in Specific Contexts
 - Chapter 5: Conclusion
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