

An Electron's Address

Name: _____

Date: _____

Hour: _____

Information: Energy Levels and Sublevels

As you know, in his solar system model Bohr proposed that electrons are located in energy levels. The current model of the atom isn't as simple as that, however.

Sublevels are located inside energy levels just like subdivisions are located inside cities. Each sublevel is given a name. Note the following table:

TABLE 1

Energy Level	Names of sublevels that exist in the energy level
1 st energy level	s
2 nd energy level	s and p
3 rd energy level	s, p, and d
4 th energy level	s, p, d, and f

Note that there is no such thing as a "d sublevel" inside of the 2nd energy level because there are only s and p sublevels inside of the 2nd energy level.

Critical Thinking Questions

1. How many sublevels exist in the 1st energy level?
2. How many sublevels exist in the 2nd energy level?
3. How many sublevels exist in the 3rd energy level?
4. How many sublevels would you expect to exist in the 5th energy level?
5. Does the 3f sublevel exist? (Note: the "3" stands for the 3rd energy level.)

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