# **IONIC BONDING**

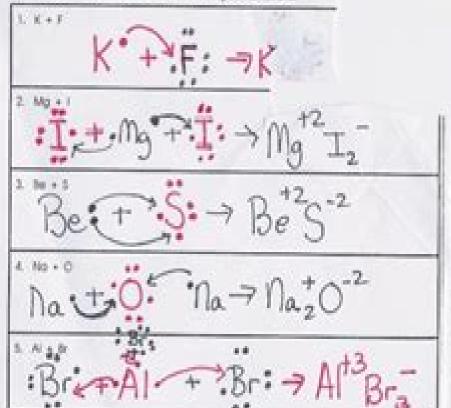
tonic bonding occurs when a metal franchis one or more electrons to a nonmetal in an effort to other a stable acted of electrons. For example, the transfer of an electron from sodium to chrome can be depicted by a Lewis dat diagram.



Colcium would need two chlorine atoms to get sid of its two voxence electrons.



Show the honder of electrons in the following combinations.



## COVALENT BONDING

Nome \_\_\_\_\_

Covalent banding occurs when two or more nonmetors share electrons, attempting to attain a stable acter of electrons at least part of the time. For example:



Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

H.+×H ->(H QH)
: F. + ^ F × → (:P. ) F:
:Ö: + ;Ö; → :Ö:;Ö:
N: + ×× × → (N(×) × × × × × × × × × × × × × × × × × ×
:O: + *C* +:O: → (O: *C*O:)
H. + X) X++H -> (HBOGH)

XX

# **Chemquest 20 Advanced Ionic Bonding Answer Key**

**Kathleen Armour** 

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