

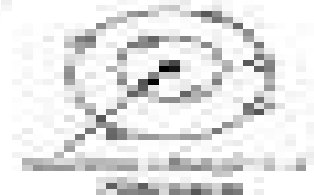
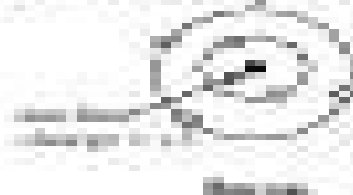
Chemistry 21



Name: Michael Kuehnle
 Date: 11/11/21
 Class: Chem

Endsemester, Abkürzung

Figure 1: "Molar Diagrams" of binary acids and oxides



Binary acids are formed by the combination of hydrogen with nonmetals. The central atom in binary acids is more electronegative than hydrogen and therefore all electrons from the hydrogen in acids are shared. Binary acids are formed by the combination of hydrogen with nonmetals. The central atom in binary acids is more electronegative than hydrogen and therefore all electrons from the hydrogen in acids are shared. Binary acids are formed by the combination of hydrogen with nonmetals. The central atom in binary acids is more electronegative than hydrogen and therefore all electrons from the hydrogen in acids are shared.

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Critical Thinking Questions

1. How large is the change in the central energy level of oxides? 100% from the original.
2. How large is the change in the central energy level of binary acids? 100% from the original.
3. How large is the change in the central energy level of oxides? 100% from the original.
4. How large is the change in the central energy level of binary acids? 100% from the original.

Chemquest 21 Answer Key

Emilie Sanchez

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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