Intro to Moles

Last Name, First Name:

Date:

Period:

Information: Atomic Mass Using Atomic Mass Units (amu) and Grams (g)

One atomic mass unit (amu) is equal to 1.6611x10<sup>-24</sup> grams.

#### Critical Thinking Questions

- According to the periodic table, a single carbon atom has a mass of 12.011 amu. What is the mass of a single carbon atom in grams?
- How many carbon atoms does it take to equal 12.011 grams?
- 3. According to the periodic table, a single phosphorus atom has a mass of 30.973 amu. What is the mass of a single phosphorus atom in grams?
- 4. How many phosphorus atoms does it take to equal 30.973 grams?
- Compare your answers to questions 2 and 4.

#### Information: What is a Mole?

Hopefully, you found that your answers to questions 2 and 4 were about the same. Both answers should be about 6.02x10<sup>23</sup>. The quantity, 6.02x10<sup>23</sup> is Avogadro's constant and we call it the "mole". Just like the quantity "12" is called a "dozen", so the quantity 6.02x10<sup>23</sup> is called a "mole".

By definition, a mole is the quantity of atoms necessary to equal the element's atomic mass in grams. So, according to the periodic table, one atom of sodium has an atomic mass of about 22.99 amu. If you weighed out 22.99 grams on a balance, you would have 6.02x10<sup>23</sup> atoms of sodium present.

Look at your periodic table and find gold (atomic number = 79). What is the mass of one gold atom? You should note that one gold atom has a mass of 196.97 amu. How many gold atoms would you need to get 196.97 grams? You would need to put 6.02 x 10<sup>23</sup> atoms of gold on the balance before you would have 196.97 grams of gold. One mole of an atom will always equal the atom's atomic mass in grams.

# **Chemquest 30 Intro To Moles Answers**

**PT Brinkman** 

**Chemquest 30 Intro To Moles Answers:** 

The book delves into Chemquest 30 Intro To Moles Answers. Chemquest 30 Intro To Moles Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemquest 30 Intro To Moles Answers, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Chemquest 30 Intro To Moles Answers
  - Chapter 2: Essential Elements of Chemquest 30 Intro To Moles Answers
  - o Chapter 3: Chemquest 30 Intro To Moles Answers in Everyday Life
  - Chapter 4: Chemquest 30 Intro To Moles Answers in Specific Contexts
  - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Chemquest 30 Intro To Moles Answers. This chapter will explore what Chemquest 30 Intro To Moles Answers is, why Chemquest 30 Intro To Moles Answers is vital, and how to effectively learn about Chemquest 30 Intro To Moles Answers.
- 3. In chapter 2, the author will delve into the foundational concepts of Chemquest 30 Intro To Moles Answers. This chapter will elucidate the essential principles that need to be understood to grasp Chemquest 30 Intro To Moles Answers in its entirety.
- 4. In chapter 3, this book will examine the practical applications of Chemquest 30 Intro To Moles Answers in daily life. This chapter will showcase real-world examples of how Chemquest 30 Intro To Moles Answers can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of Chemquest 30 Intro To Moles Answers in specific contexts. The fourth chapter will explore how Chemquest 30 Intro To Moles Answers is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Chemquest 30 Intro To Moles Answers. This chapter will summarize the key points that have been discussed throughout the book.
  - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemquest 30 Intro To Moles Answers.

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