

2021 WASSCE CHEMISTRY THEORY

1.

(a) Distinguish between molecular formula and structural formula.

(b) List three factors that determine the ionization energy of an atom.

(c) State two conditions necessary for the establishment of chemical equilibrium.

(d) Consider the following table.

<i>Element</i>	A	B	C
<i>Ionization energy kJ/mol</i>	619	518	594

(i) State which of the elements is the strongest reducing agent.

(ii) Give the reason for the answer stated in 1(d)(i).

(e) State Graham's law of diffusion.

(f) Consider the following salts:

$\text{Mg}(\text{NO}_3)_2$, CaCO_3 , Na_2SO_4 .

State which of the salts is/are:

(i) readily soluble in water

(ii) insoluble in water

(g) Classify each of the following products as addition polymer or condensation polymer:

(i) Protein

(ii) Perspex

(iii) Nylon

(h) Define atomic radius

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S Ashworth



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Table of Contents Chemistry Theory Paper21april Answer

1. Understanding the eBook Chemistry Theory Paper21april Answer
 - The Rise of Digital Reading Chemistry Theory Paper21april Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Theory Paper21april Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Theory Paper21april Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Theory Paper21april Answer
 - Personalized Recommendations
 - Chemistry Theory Paper21april Answer User Reviews and Ratings
 - Chemistry Theory Paper21april Answer and Bestseller Lists
5. Accessing Chemistry Theory Paper21april Answer Free and Paid eBooks
 - Chemistry Theory Paper21april Answer Public Domain eBooks
 - Chemistry Theory Paper21april Answer eBook Subscription Services
 - Chemistry Theory Paper21april Answer Budget-Friendly Options
6. Navigating Chemistry Theory Paper21april Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Theory Paper21april Answer Compatibility with Devices
 - Chemistry Theory Paper21april Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Theory Paper21april Answer
 - Highlighting and Note-Taking Chemistry Theory Paper21april Answer
 - Interactive Elements Chemistry Theory Paper21april Answer

8. Staying Engaged with Chemistry Theory Paper21april Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Theory Paper21april Answer
9. Balancing eBooks and Physical Books Chemistry Theory Paper21april Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Theory Paper21april Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Theory Paper21april Answer
 - Setting Reading Goals Chemistry Theory Paper21april Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Theory Paper21april Answer
 - Fact-Checking eBook Content of Chemistry Theory Paper21april Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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