Chemistry – Unit 1 - Worksheet 6

Dimensional Analysis

Use the factor-label method to make the following conversions. Remember to use the appropriate number of sf's in your answer.

Part 1
1. 74 cm x
$$\frac{1}{100}$$
 cm $= \frac{0.74}{100}$ meters

2. 8.32 x 10-2 kg x 1000 g = 057 grams

3. 55.5 mL x 1 cm3 = 55.5 cm3

4. 0.00527 cal x 1 kilocalories =
$$5.27 \times 10^{-6}$$
 kilocalories

5.
$$9.52 \times 10^{-4} \text{ m} \times \frac{10^{6} \text{ jum}}{10^{6}} = \frac{952}{10^{6}}$$
 micrometers

7.
$$6.0 \times 10^{-1} \text{ g} \times \frac{100009}{15} = \frac{6.0 \times 10^{3} \text{ mg}}{15}$$
8. $8.34 \times 10^{-9} \text{ cg} \times \frac{19}{10009} = \frac{8.34 \times 10^{-11}}{5.0}$

8. 8.34 x 10-9 cg x
$$\frac{19}{100000} = \frac{8.34 \times 10^{-1}}{9}$$

12.
$$9.1 \times 10^{-13} \text{ kg.} \times \frac{10009}{1 \text{ Kg}} \times \frac{10^9 \text{ rg}}{1 \text{ g}} = 0.91 \text{ rg}$$

Chemistry Unit Size Of Things Answers

Marilyn Jager Adams

Chemistry Unit Size Of Things Answers:

Unveiling the Magic of Words: A Report on "Chemistry Unit Size Of Things Answers"

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