

Covalent Bonding Study Guide

Use the book and notes to answer these questions. You will need to check your answers on Edmodo before you turn this in.

- How many valence electrons are in the following:
 - Carbon 4
 - Silicon 4
 - Chlorine 7
 - Nitrogen 5
- What is the name given to the pairs of valence electrons that do not participate in bonding?
Lone or unpaired
- What is the name given to the pairs of valence electrons that do participate in bonding?
Shared pair
- In a covalent bond, electrons are shared between atoms.
- Which of the following is a covalent molecule (circle one)?
 - LiCl
 - BaO
 - NO₂
 - MgBr₂

Writing Formulas for Covalent Compounds

- Write the formula for the following covalent compounds:
 - Sulfur tribromide *SBr₃*
 - Dinitrogen tetrachloride *N₂Cl₄*
 - Dihydrogen monoxide *H₂O*
 - Phosphorus Trihydride *PH₃*
 - Tetracarbon octahydride *C₄H₈*

Nomenclature for Covalent Compounds

- Name the following covalent compounds:
 - N₄O₆ *tetranitrogen hexoxide*
 - SO₃ *sulfur trioxide*
 - SeF₄ *selenium tetrafluoride*
 - PCl₅ *phosphorous pentachloride*
 - XeI₄ *xenon tetraiodide*

Prefix	Meaning
Mono-	1
Di-	2
Tri-	3
Tetra-	4
Penta-	5
Hexa-	6
Hepta-	7
Octa-	8
Nona-	9
Deca-	10

Chemistry Unit 6 Study Guide Covalent Bonding

A Loxley



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