Mole Problems Worksheet

Balance the following equations and answer the respective questions.

1. _____ NO + _____ O₂ → _____ NO₂

- (a) 2 moles of NO will react with _____ mole(s) of O2 to produce _____ mole(s) of NO2.
- (b) moles NO₂ = 3.6 moles O₂ × (2 moles of NO₂/ 1 mole of O₃)
- (c) Approximately mol NO is required to produce 4.67 mol NO,

2. _____ AIF₃ + _____ O₂ → _____ AI₂O₃ + _____ F₂

- (a) 20 moles of AIF₃ will produce moles of F₃.
- (b) _____ moles of AIF₃ will react with 0.6 moles of O₂.

O₃ + Fe → Fe₃O₃

- (a) ____ moles of oxygen gas react with ____ moles of iron to produce ____ moles of iron (III) oxide.
- (b) _____ moles of O₂ are required to produce 3.0 moles of iron (III) oxide.

4. _____ NH₃ + _____ O₂ → _____ N₂ + _____ H₂O

- (a) 20 moles of NH₃ are needed to produce _____ moles of H₂O.
- (b) _____ mol N₂ is produced from 3.5 mol O₂ in this reaction.

<u>Chemistry Worksheet 7 General Review Of Mole Problems</u>

SJ Ball

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