

Intro. to Chemical Reactions

Name _____

Date _____

Section _____

Formula Review

1. Write the formulas for the following compounds:

- a. aluminum phosphate _____ b. sodium phosphate _____
 c. copper(II) phosphate _____ d. sodium sulfate _____

2. Write the formulas for the following compounds:

- a. CaCl_2 _____ b. FeCl_3 _____
 c. FeCl_2 _____ d. CuCl_2 _____

Chemical Reactions: Virtual Energy Table

Write down observations for the reaction as it occurs:



There is no observable gas or color change when the two solutions are mixed.

Experiment 1 involves the reaction of hydrochloric acid and sodium hydroxide (also known as NaOH).

Write a balanced chemical equation for the reaction between sodium hydroxide and hydrochloric acid. Label the products.

Chemical Table Data Collection

3. Write the following reaction under the correct heat flow description:



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