

EXERCISE NO. 4

DENSITY, SPECIFIC GRAVITY AND SPECIFIC VOLUME

I. INTENDED LEARNING OUTCOMES

At the end of the exercise, the students should be able to:

1. calculate density, specific gravity and specific volume of different liquids.
2. determine the specific gravity of unknown liquids using the data obtained from the specific gravity bottle (pycnometer) method.
3. calculate weight or volume of the substance given the specific gravity.

II. DISCUSSION

Density is defined as mass per unit volume of a substance at a fixed temperature and pressure. Density can be calculated by dividing the mass of a substance by its volume, expressed as: $Density = \frac{Mass}{Volume}$. The density of liquids is usually expressed as grams per milliliter (g/mL).

Specific gravity is the ratio of the weight of a given substance to the weight of an equal volume of water at a similar temperature, expressed as:

$$\text{Specific gravity} = \frac{\text{Weight of substance}}{\text{Weight of equal volume of water}}$$

In the U.S. Pharmacopeia, the standard temperature for specific gravities is 25°C, unless otherwise stated. Specific gravity is a dimensionless property and therefore has no units. The specific gravity of water is 1; therefore, the volume of water in milliliters is usually equal to its weight in grams. If the weight of a particular volume of a substance is known, its specific gravity can be calculated by dividing its weight by its volume:

$$\text{Specific gravity} = \frac{\text{Weight of substance (g)}}{\text{Volume of substance (mL)}}$$

This equation is similar to the one for density. So, an easy way to calculate specific gravity is to calculate the density and write the value without any units. If the specific gravity of a solution is known, interconversions between the volume of a substance and its weight are possible using the following expressions:

$$\text{Weight of substance (g)} = \text{Volume of substance (mL)} \times \text{Specific gravity}$$

$$\text{Volume of substance (mL)} = \frac{\text{Weight of substance (g)}}{\text{Specific gravity}}$$

Density And Specific Gravity Laboratory Manual

William Suddards Franklin



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