1. A compound contains 36.5% Na, 25.4% S, and 38.1% O. Find its empirical formula.

2. Find the empirical formula of a compound with 53.7% iron and 46.3% sulfur.

Analysis of a compound sample indicates that it has 1.04 g K, 0.70 g Cr, and 0.86 g O. What is its empirical formula?

4. An acid commonly used in the automotive industry is shown to be 31.6% phosphorous, 3.1% hydrogen, and 63.5% oxygen. Determine the empirical formula of this acid.

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**Dale Stroble Davis** 

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