



# Determining Empirical Formula Worksheet



1. A solvent is found to be 50.0% oxygen, 37.5% carbon, and 12.5% hydrogen. What is the empirical formula of this solvent?
2. A particular sugar is determined to have the following composition: 40.0% carbon, 6.7% hydrogen, and 53.5% oxygen. Determine the empirical formula of this sugar molecule.
3. If 4.04 g of nitrogen combine with 11.46 g of oxygen to produce a compound with a molar mass of 108.0 g, what is the molecular formula of this compound?
4. The molar mass of a compound is 92 g. Analysis of the sample indicates that it contains 0.606 g N and 1.390 g O. Find the compound's molecular formula.

# Determining Empirical Formulas Worksheet Answers Instructional Fair

**Theodore Rudolph Running**



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*Chemistry Homework* Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more *Resources in Education*, 1984 Experiment to Determine Relative Values of Graphic Formulas as Contrasted with Empirical Formulas in Teaching of Elementary Chemistry George W. Richards, 1926 **Empirical Formulas** Theodore Rudolph Running, 1917 Solving Empirical Formula Problems, 2016 Continue your consideration of classic chemistry problems with a look at empirical formulas and examine how empirical formulas relate to molecular formulas *Empirical Formulas* HardPress, Running Theodore R 1866, 2013-01 Unlike some other reproductions of classic texts 1 We have not used OCR Optical Character Recognition as this leads to bad quality books with introduced typos 2 In books where there are images such as portraits maps sketches etc We have endeavoured to keep the quality of these images so they represent accurately the original artefact Although occasionally there may be certain imperfections with these old texts we feel they deserve to be made available for future generations to enjoy Empirical Formulas Th.R. Running, 1917 Empirical Formulas Theodore Rudolph Running, 1917 **Empirical Formulas** Runni Theodore R (Theodore Rudolph), 2022-10-27 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant *Percent Composition and Empirical Formulas*, 2016 In this lecture you encounter two classic types of chemistry problems and learn the basic characteristics of each The lecture concludes with several practice problems to help you master the skill of solving percent composition problems *Empirical Formulas* Theodore Rudolph Running, 2006

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