

11

CHEMICAL REACTIONS

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

► Writing Chemical Equations (pages 321–323)

1. A chemical reaction occurs when one or more _____ change into one or more new substances called _____.
2. The arrow in a reaction means _____.
3. Is the following sentence true or false? When there are two or more reactants or products, they are separated by an arrow. _____
4. Write a word equation that describes the following reactions.
 - a. Acetylene reacts with oxygen to produce carbon dioxide and water.

 - b. When heated, mercury(II) oxide reacts to form mercury and oxygen.

5. What is a chemical equation?

6. A chemical reaction that shows only the formulas, but not the relative amounts of the reactants and products is a(n) _____.
7. Identify the reactant(s) and product(s) in the chemical equation $\text{Li} + \text{Br}_2 \longrightarrow \text{LiBr}$.
 - a. reactant(s) _____
 - b. product(s) _____
8. Circle the letter of each statement that is true about a catalyst.
 - a. A catalyst is the new material produced as a result of a chemical reaction.
 - b. A catalyst is not used up in a chemical reaction.
 - c. A catalyst adds heat to a chemical reaction.
 - d. A catalyst speeds up a chemical reaction.

Describing Chemical Change Answers

Michael Hoelscher



Describing Chemical Change Answers:

A History of Chemical Theories and Laws Matthew Moncrieff Pattison Muir, 1907 **Cambridge International AS/A Level Chemistry Revision Guide 2nd edition** David Bevan, 2016-01-25 Get your best grades with this exam focused text that will guide you through the content and skills you need to prepare for the big day Manage your own revision with step by step support from experienced examiner and author David Bevan This guide also includes a Questions and Answers section with exam style questions student s answers for each question and examiner comments to ensure you re exam ready Plan and pace your revision with the revision planner Use the expert tips to clarify key points Avoid making typical mistakes with expert advice Test yourself with end of topic questions and answers and tick off each topic as you complete it Practise your exam skills with exam style questions and answers This title has not been through the Cambridge International endorsement process **Philosophies of the Sciences** Fritz Allhoff, 2015-12-02 A collection of essays discussing a wide range of sciences and the central philosophical issues associated with them presenting the sciences collectively to encourage a greater understanding of their associative theoretical foundations as well as their relationships to each other Offers a new and unique approach to studying and comparing the philosophies of a variety of scientific disciplines Explores a wide variety of individual sciences including mathematics physics chemistry biology psychology sociology and economics The essays are written by leading scholars in a highly accessible style for the student audience Complements more traditional studies of philosophy of science Water for life The Open University, This free 15 hour course explained how the basic chemistry of how atoms elements and molecules work together **Introduction to General Inorganic Chemistry** Alexander Smith, 1906 Designs for Science Literacy American Association for the Advancement of Science, 2001-03-22 The call for science curriculum reform has been made over and over again for much of the twentieth century Arguments have been made that the content of the curriculum is not appropriate for meeting the individual and social needs of people living in the modern world that the curriculum has become overstuffed with topics and does not serve students especially well and above all that the curriculum does not generate the student learning it is expected to produce The latest volume in a continuing series of publications from the AAAS designed to reform science education Designs for Science Literacy presupposes that curriculum reform must be considerably more extensive and fundamental than the tinkering with individual courses and subjects that has been going on for decades Designs deals with the critical issues involved in assembling sound instructional materials into a new coherent K 12 whole The book pays special attention to the need to link science oriented studies to the arts and humanities and also proposes how to align the curriculum with an established set of learning goals while preserving the American tradition of local responsibility for the curriculum itself If fundamental curriculum reform is ever to occur a new process for creating alternatives will have to be developed Designs for Science Literacy provides the groundwork for such a process **Addison-Wesley Chemistry** Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L.

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