

20.1**DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

Key Terms

- acid
- base

Part A Completion

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of _____ 1 _____ are electrolytes. Acids react with many _____ 2 _____ to produce _____ 3 _____ gas. _____ 4 _____ are compounds that react with acids to form _____ 5 _____ and a salt.

Acids produce _____ 6 _____ ions when dissolved in water. There is a standard system for naming acids. For example, HBr is called _____ 7 _____ acid, while HNO_3 is called _____ 8 _____ acid.

Bases produce _____ 9 _____ ions when dissolved in water. Some compounds that are _____ 10 _____ are named in the same way as other ionic compounds.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

Part B True-False

Circlely write *if these classes are true* (abbreviate as *AC*) *some classes* (true), *SC*, *no answer* (true), *NT*.

- _____ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- _____ 12. An acid contains one or more hydroxide ions.
- _____ 13. Chemists have a special system for naming bases.

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Table of Contents Describing Acids And Bases 20 1

1. Understanding the eBook Describing Acids And Bases 20 1
 - The Rise of Digital Reading Describing Acids And Bases 20 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Describing Acids And Bases 20 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Describing Acids And Bases 20 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Describing Acids And Bases 20 1
 - Personalized Recommendations
 - Describing Acids And Bases 20 1 User Reviews and Ratings
 - Describing Acids And Bases 20 1 and Bestseller Lists
5. Accessing Describing Acids And Bases 20 1 Free and Paid eBooks
 - Describing Acids And Bases 20 1 Public Domain eBooks
 - Describing Acids And Bases 20 1 eBook Subscription Services
 - Describing Acids And Bases 20 1 Budget-Friendly Options
6. Navigating Describing Acids And Bases 20 1 eBook Formats

- ePub, PDF, MOBI, and More
- Describing Acids And Bases 20 1 Compatibility with Devices
- Describing Acids And Bases 20 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Describing Acids And Bases 20 1
 - Highlighting and Note-Taking Describing Acids And Bases 20 1
 - Interactive Elements Describing Acids And Bases 20 1
- 8. Staying Engaged with Describing Acids And Bases 20 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Describing Acids And Bases 20 1
- 9. Balancing eBooks and Physical Books Describing Acids And Bases 20 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Describing Acids And Bases 20 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Describing Acids And Bases 20 1
 - Setting Reading Goals Describing Acids And Bases 20 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Describing Acids And Bases 20 1
 - Fact-Checking eBook Content of Describing Acids And Bases 20 1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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