

$$.777g \text{ H}_2\text{O} \frac{1 \text{ mol H}_2\text{O}}{18.02g \text{ H}_2\text{O}} = .0431$$

0.0431 moles  $\text{H}_2\text{O}$

2. Calculate the moles of copper in your sample of copper chloride?

$$.742g \text{ Cu} \frac{1 \text{ mol Cu}}{63.546g \text{ Cu}} = 0.0117$$

0.0117 moles Cu

3. Calculate the moles of chlorine in your sample of copper chloride?

$$M_{\text{Cl}} = M_{\text{Cu}} - M_{\text{hydrate}} = 0.7415 - 0.227 = .515g \text{ Cl}$$

$$.515g \text{ Cl} \frac{1 \text{ mol Cl}}{35.4527} = 0.0145$$

0.0145 moles Cl

4. Write the proper chemical formula for the compound that you tested.

$$\text{Moles Cu} = 0.0117 / 0.0117 = 1.45 = 5$$

$$\text{Moles Cl} = 0.0145 / 0.0117 = 1.245 = 6$$



Correct formula for initial hydrated salt



5. Explain using chemical equations the process of a) dehydrating the hydrate salt b) dissolving the dehydrated salt c) the change in copper observed after adding the aluminum wire to the solution.

# Determination Of A Chemical Formula Answers

**Burdge**



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**A Textbook of Organic Chemistry, 22e** Arun Bahl & B S Bahl, 2019 With an increased focus on fundamentals this new edition of A Textbook of Organic Chemistry continues to present the time tested functional group approach to the subject This examination oriented book breaks the intricacies of Organic Chemistry into easy to understand steps which gives the student the necessary foundation to build upon learn and understand Organic Chemistry in a way that is efficient as well as long lasting

**CBSE Chemistry Chapterwise Case Study Class 11** Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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**Basics for Chemistry** David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the

study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material     *Descriptive Chemistry* Lyman Churchill Newell,1903     *General Chemistry ...* Lyman Churchill Newell,1914

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