# **Hints to Balancing Chemical Equations**

- Adjust the coefficients not the subscripts.
- Even/Odd Multiply by 2 to make all even.
- Balance polyatomic ions as a whole.
- Check the balanced equation.

# Example:

$$ZnS + O_2 \Rightarrow ZnO + SO_2$$
 (O is even on the left and odd on the right)

$$ZnS + O_2 \Rightarrow 2ZnO + SO_2$$
 (multiply by 2 to make O even)

# Example:

# **Balance Word Equations Chemistry**

Earl Rouse Glenn, Louis E. Welton

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A Concrete Stoichiometry Unit for High School Chemistry Jennifer Louise Pakkala,2006 Chemical Symbolism and Calculations Stanley Wallace Morse,1953 Holt Chemistry Ralph Thomas Myers,2004 Chemistry of Hazardous Materials Eugene Meyer,1989

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