Block

```
    Zn + 2 HCl → ZnCl<sub>2</sub> + H<sub>2</sub>

    4 Fe + 13 O2 - 2 Fe2O3
3. L_4 P + 5 O_2 \rightarrow P_4 O_{10}
   ____ N2 + _Z_ H2 -> _ 2 NH3
   ____ PbO<sub>2</sub> -> ____ Pb + ____ O<sub>2</sub>
   2 CoHo+ 15 O2 - 12 CO2+ La H2O
   ____ C2H3OH + 3 O2 -> 2 CO2 + 3 H2O
7.
     3 Fe+ 4 H2O - Fe3O4+ 4 H2
     ____ SiO2 + 4 HF → ____ SiF4 + 2 H2O
10. 3 CO+ FeO4 → FeO+ 3 CO2
11. NH_4NO_2 \rightarrow N_2 + 2 H_2O
         N2H4+ ___ O2 -> ___ N2+_2 H2O
12.
      3 NO+ H2O → 2 HNO+ NO
13.
C12H22O1+ 12 O2 - 12 CO2+ 11 H20
        NH + 3 0 - 6 H20 + 2 No
    _{--} CH<sub>3</sub>COOH + _{--} O<sub>2</sub> \rightarrow _{-} \bigcirc CO<sub>2</sub> + _{-} \bigcirc H<sub>2</sub>O
17.
    2 C2H6+ 7 O2 - 4 CO2+ 6 H2O
19. ____ C<sub>0</sub>H<sub>8</sub> + 10 O<sub>2</sub> → 8 CO<sub>2</sub> + 4 H<sub>2</sub>O
20. 2 C_4H_{10} + 13 O_2 \rightarrow 8 CO_2 + 10 H_2O
21. N_2 + 3 H_2 \rightarrow 2 NH_3
22. 3 Fe+ 4 H2O - Fe3O4+ 4 H2
23. 3 Fe+ 2-02 → ___ Fe3O4
      3 NH4OH+ ___ AICI3 → ___ AI(OH)3 + _ 3 NH4CI
25. Na_2CO_3 + \underline{\hspace{1cm}} H_2O + \underline{\hspace{1cm}} CO_2 \rightarrow \underline{\hspace{1cm}} RAHCO_3
        MnOz+ 4 HCl → ___ MnClz + 2 HzO + ___ Clz
26.
        _PBr3 + 3 H2O → ___ H3PO3 + 3 HBr
27. ___
28. 2 ZnS+ 3 O₂ → 2 ZnO+ 5O₂
29. 2 NH3+ 3 O2 → N2+ 6 H2O
         N_{02}B_4O_7 + ___ H_2SO_4 + __ 5 H_2O \rightarrow ___ N_{02}SO_4 + _4 H_3BO_3
30. ___
      2 NH3+ 2 02 → ____NO+ 3 H2O
31.
      4 HNO3 - 2 H2O+ 4 NO2+ 02
32. _
      4 HCI+ 2 HNO3 -> 2 CI2+ 2 NOCI+ 3 HO
33. _
34. 2 Ca3(PO4)2+ 6 SIO2 - 6 CaSiO3 + P4O10
35. _____ C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> → 2 C<sub>2</sub>H<sub>6</sub>OH + 2 CO<sub>2</sub>
36. _2 CBH18+25 O2 -> 16 CO2+ 18 HO
37. 2 CH4+ 3 02 → 2 CO+ 4 H2O
38. Q C_2H_2 + 3 O_2 \rightarrow 4 CO_2 + 2 H_2O
39. b HoO+ b CO2 → ____ CoH 200 + b O2
40. ____ Cu + _ U + _ HNO_3 \rightarrow _ Cu(NO_3)_2 + _ NO + _ H_2O
```

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Peipei Pang

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- 1. Understanding the eBook Balancing Equations 3 Answer Sheet
 - The Rise of Digital Reading Balancing Equations 3 Answer Sheet
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Balancing Equations 3 Answer Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Balancing Equations 3 Answer Sheet
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Balancing Equations 3 Answer Sheet
 - Personalized Recommendations
 - Balancing Equations 3 Answer Sheet User Reviews and Ratings
 - Balancing Equations 3 Answer Sheet and Bestseller Lists
- 5. Accessing Balancing Equations 3 Answer Sheet Free and Paid eBooks

- Balancing Equations 3 Answer Sheet Public Domain eBooks
- Balancing Equations 3 Answer Sheet eBook Subscription Services
- Balancing Equations 3 Answer Sheet Budget-Friendly Options
- 6. Navigating Balancing Equations 3 Answer Sheet eBook Formats
 - o ePub, PDF, MOBI, and More
 - Balancing Equations 3 Answer Sheet Compatibility with Devices
 - Balancing Equations 3 Answer Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Balancing Equations 3 Answer Sheet
 - Highlighting and Note-Taking Balancing Equations 3 Answer Sheet
 - Interactive Elements Balancing Equations 3 Answer Sheet
- 8. Staying Engaged with Balancing Equations 3 Answer Sheet
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Balancing Equations 3 Answer Sheet
- 9. Balancing eBooks and Physical Books Balancing Equations 3 Answer Sheet
 - Benefits of a Digital Library
 - $\circ\,$ Creating a Diverse Reading Collection Balancing Equations 3 Answer Sheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Balancing Equations 3 Answer Sheet
 - Setting Reading Goals Balancing Equations 3 Answer Sheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Balancing Equations 3 Answer Sheet
 - Fact-Checking eBook Content of Balancing Equations 3 Answer Sheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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