# Alpha (α) Radiation

$$^{238}_{92}U \longrightarrow ^{234}_{90}Th + ^{4}_{2}He$$

Mass number: 
$$238 = 234 + 4$$

Atomic number: 
$$92 = 90 + 2$$

# **Balancing Of Radioactive Equation**

**Sir Arthur Stanley Eddington** 

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