

Balancing Chemical Equations Worksheet

1. _____ H_2 + _____ $\text{O}_2 \rightarrow$ _____ H_2O
2. _____ N_2 + _____ $\text{H}_2 \rightarrow$ _____ NH_3
3. _____ S_8 + _____ $\text{O}_2 \rightarrow$ _____ SO_3
4. _____ N_2 + _____ $\text{O}_2 \rightarrow$ _____ N_2O
5. _____ $\text{HgO} \rightarrow$ _____ Hg + _____ O_2
6. _____ CO_2 + _____ $\text{H}_2\text{O} \rightarrow$ _____ $\text{C}_6\text{H}_{12}\text{O}_6$ + _____ O_2
7. _____ Zn + _____ $\text{HCl} \rightarrow$ _____ ZnCl_2 + _____ H_2
8. _____ SiCl_4 + _____ $\text{H}_2\text{O} \rightarrow$ _____ H_4SiO_4 + _____ HCl
9. _____ Na + _____ $\text{H}_2\text{O} \rightarrow$ _____ NaOH + _____ H_2
10. _____ $\text{H}_3\text{PO}_4 \rightarrow$ _____ $\text{H}_4\text{P}_2\text{O}_7$ + _____ H_2O
11. _____ $\text{C}_{10}\text{H}_{16}$ + _____ $\text{Cl}_2 \rightarrow$ _____ C + _____ HCl
12. _____ CO_2 + _____ $\text{NH}_3 \rightarrow$ _____ $\text{OC}(\text{NH}_2)_2$ + _____ H_2O
13. _____ Si_2H_5 + _____ $\text{O}_2 \rightarrow$ _____ SiO_2 + _____ H_2O_3
14. _____ $\text{Al}(\text{OH})_3$ + _____ $\text{H}_2\text{SO}_4 \rightarrow$ _____ $\text{Al}_2(\text{SO}_4)_3$ + _____ H_2O
15. _____ Fe + _____ $\text{O}_2 \rightarrow$ _____ Fe_2O_3
16. _____ $\text{Fe}_2(\text{SO}_4)_3$ + _____ $\text{KOH} \rightarrow$ _____ K_2SO_4 + _____ $\text{Fe}(\text{OH})_3$
17. _____ $\text{C}_7\text{H}_6\text{O}_2$ + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O
18. _____ H_2SO_4 + _____ $\text{HI} \rightarrow$ _____ H_2S + _____ I_2 + _____ H_2O
19. _____ FeS_2 + _____ $\text{O}_2 \rightarrow$ _____ Fe_2O_3 + _____ SO_2
20. _____ Al + _____ $\text{FeO} \rightarrow$ _____ Al_2O_3 + _____ Fe
21. _____ Fe_2O_3 + _____ $\text{H}_2 \rightarrow$ _____ Fe + _____ H_2O
22. _____ Na_2CO_3 + _____ $\text{HCl} \rightarrow$ _____ NaCl + _____ H_2O + _____ CO_2
23. _____ K + _____ $\text{Br}_2 \rightarrow$ _____ KBr
24. _____ C_7H_{16} + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O
25. _____ P_4 + _____ $\text{O}_2 \rightarrow$ _____ P_2O_5

Balancing Chemical Equations 1 50

Jianjun Gao



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