Name:



Balancing Chemical Equations

Directions: Balance each of the following chemical equations.

- 1) $C_{10}H_{22} + C_{02} \rightarrow C_{02} + C_{02}H_{20}$
- 2) ___ Al(OH)₃ + ___ HBr → ___ AlBr₃ + ___ H₂O
- 3) ___CH₂CH₂CH₂CH₃ + ___O₂ → ___CO₂ + ___H₂O
- 4) __C₃H₈ + __O₂ → __CO₂ + __H₂O
- 5) __ Li + __ AlO₃ → __ LICI + __ Al
- 6) ___C₂H₆ + ___O₂ → ___CO₂ + ___H₂O
- 7) ___NH4OH + ___H2O4 → ___(NH4)3PO4 + ___H2O
- 8) ___ Rb + ___ P → ___ Rb₃P
- 9) __CH₄ + __O₂ → __CO₂ + __H₂O
- 10) ___ Al(OH)₃ + ___ H₂SO₄ → ___ Al₂(SO₄)₃ + ___ H₂O
- 11) ___ Na + ___ Cl₂ → ___ NaCl
- 12) __ Rb + __ S₈ → __ Rb₂S
- 13) ___ H_3PO_4 + ___ $Ca(OH)_2$ \rightarrow ___ $Ca_3(PO_4)_2$ + ___ H_2O
- 14) ___ NH₃ + ___ HCl → ___ NH₄Cl
- 15) __ Li + __ H₂O → __ LiOH + __ H₂
- 16) ___ $Ca_3(PO_4)_2 +$ ___ $SIO_2 +$ ___ $C \rightarrow$ ___ $CaSIO_3 +$ ___ CO + ___ P
- 17) ___NH₃ + ___O₂ → ___N₂ + ___H₂O
- 18) ___ FeS₂ + ___ O₂ → ___ Fe₂O₃ + ___ SO₂

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J Dewey

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