

Balancing Equations Worksheet – Answers

Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number “1”.

- 1) $1 \text{ Na}_3\text{PO}_4 + 3 \text{ KOH} \rightarrow 3 \text{ NaOH} + 1 \text{ K}_3\text{PO}_4$
- 2) $1 \text{ MgF}_2 + 1 \text{ Li}_2\text{CO}_3 \rightarrow 1 \text{ MgCO}_3 + 2 \text{ LiF}$
- 3) $1 \text{ P}_4 + 3 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_3$
- 4) $2 \text{ RbNO}_3 + 1 \text{ BeF}_2 \rightarrow 1 \text{ Be}(\text{NO}_3)_2 + 2 \text{ RbF}$
- 5) $2 \text{ AgNO}_3 + 1 \text{ Cu} \rightarrow 1 \text{ Cu}(\text{NO}_3)_2 + 2 \text{ Ag}$
- 6) $1 \text{ CF}_4 + 2 \text{ Br}_2 \rightarrow 1 \text{ CBr}_4 + 2 \text{ F}_2$
- 7) $2 \text{ HCN} + 1 \text{ CuSO}_4 \rightarrow 1 \text{ H}_2\text{SO}_4 + 1 \text{ Cu}(\text{CN})_2$
- 8) $1 \text{ GaF}_3 + 3 \text{ Cs} \rightarrow 3 \text{ CsF} + 1 \text{ Ga}$
- 9) $1 \text{ BaS} + 1 \text{ PtF}_2 \rightarrow 1 \text{ BaF}_2 + 1 \text{ PtS}$
- 10) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$
- 11) $2 \text{ NaF} + 1 \text{ Br}_2 \rightarrow 2 \text{ NaBr} + 1 \text{ F}_2$
- 12) $1 \text{ Pb}(\text{OH})_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$
- 13) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$
- 14) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$
- 15) $2 \text{ Na}_3\text{PO}_4 + 3 \text{ CaCl}_2 \rightarrow 6 \text{ NaCl} + 1 \text{ Ca}_3(\text{PO}_4)_2$
- 16) $2 \text{ K} + 1 \text{ Cl}_2 \rightarrow 2 \text{ KCl}$
- 17) $2 \text{ Al} + 6 \text{ HCl} \rightarrow 3 \text{ H}_2 + 2 \text{ AlCl}_3$
- 18) $1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$
- 19) $1 \text{ SO}_2 + 2 \text{ Li}_2\text{Se} \rightarrow 1 \text{ SSe}_2 + 2 \text{ Li}_2\text{O}$
- 20) $2 \text{ NH}_3 + 1 \text{ H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$

Balancing Chemical Equations Chapter 7 Worksheet

Word Equations

**Arthur E. Morris, H. Alan Fine, Gordon
Harold Geiger**



Balancing Chemical Equations Chapter 7 Worksheet Word Equations:

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study.

Features
Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule.

Workflow
Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

About the Author
DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

Learning Elementary Chemistry for Class 7 (A.Y. 2023-24) Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum. The content of this series has been developed to fulfill the requirement of all the six domains: Concepts, Processes, Applications, Attitudes, Creativity, and World view of Science, to make teaching and learning of Chemistry interesting, understandable, and enjoyable for young minds. This series builds a solid foundation for young learners to prepare them for higher classes. The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry. Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures. Unique features of this series are: The content of the book is written in a very simple and easy-to-understand language. All the

Key concepts in the curriculum have been systematically covered and graded in the text. Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit. At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given. Explanations, illustrations, diagrams, experiments and solutions to numerical problems have been included to make the subject more interesting, comprehensive and appealing. Diagrams, illustrations and text have been integrated to enhance comprehension. Definitions and other important scientific information are highlighted. Throughout the series, investigations related to the text enable the learners to learn through experimentation. Quick revision of each chapter has been given under the caption Highlights in Review. Online Support: It provides Video lectures, Unit wise interactive exercises, Chapterwise Worksheet, Solution of textbook questions for Teachers only, E Book for Teachers only. I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum. Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book.

Author **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day. Relevant content, lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them. Now includes even more technology tools and activities to support differentiated instruction. *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM** Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger, 2011-09-06. This book approaches the subject of material and energy balances from two directions. First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles. Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances. The third edition reflects the changes in the professional engineer's practice in the last 30 years reflecting the dramatic shift away from metallurgical engineering and the extractive industry towards materials engineering. A large and growing number of recent graduates are employed in such fields as semiconductor processing, environmental engineering and the production and processing of advanced and exotic materials for aerospace, electronic and structural applications. The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems. The spreadsheet program Excel is used extensively throughout the text as the main computational engine for solving material and energy balance equations and for statistical analysis of data. The use of Excel and the introduction of the add-in programs enables the study of a range of variables on critical process parameters.

and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and units **Discover Science: Teacher's annotated edition** ,1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today s technological society Objective Workbook for Simplified ICSE Chemistry , **Glencoe Science** McGraw-Hill Staff,2001-06 **Te HS&T a** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02 **Te HS&T J** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02

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