

Balancing Chemical Equations

In any chemical reaction, the chemicals that are mixed together are called the reactants. The reactants are written to the left of the arrow. The chemicals that are produced in the reaction are called the products.



In the above reaction we see that one molecule of H_2 reacts with one molecule of O_2 to make one molecule of H_2O . Let's use the following table to summarize the total number of atoms on each side of the equation:

Reactants	Products
2 - H's	2 - H's
2 - O's	1 - O

This equation is not "balanced" at the present time because there are different number of oxygen atoms on each side of the arrow. We can balance the equation however, by adding some coefficients.



Notice what we have done. We have equaled out the number of hydrogen and oxygen atoms on each side of the equation. Let's look at a summary of the atoms present now.

Reactants	Products
4 - H's	4 - H's
2 - O's	2 - O's

Equations such as this one, in which the number of each of the individual atoms are equal on both sides of the equation, are said to be balanced.

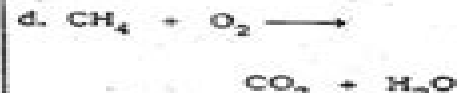
One key point that we need to stress when we work on balancing chemical equations is that we only change the coefficients; never the subscripts, in order to achieve equal numbers of atoms on each side.



The 5 is called a coefficient; and coefficients can be changed to balance reactions. Changing a coefficient simply changes the number of molecules present, but does not affect the composition of the molecules.

The 2 & 4 are called subscripts; and subscripts can't be changed in order to balance an equation. Changing any of the subscripts would result in a new compound which we do not want.

Let's take a few minutes and see if we can balance a few practice equations together.



So far, we have balanced our equations by "inspection." In other words simply by a combination of logic and trial-and-error. This is usually the fastest method to use, however, as you will see later in the section on oxidation and reduction, there are systematic methods to use that can help you balance equations.

Let's take a few minutes to learn a few tricks that might help you balance some equations...

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 - The Rise of Digital Reading Balancing Equations Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Balancing Equations Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Balancing Equations Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Balancing Equations Packet
 - Personalized Recommendations
 - Balancing Equations Packet User Reviews and Ratings
 - Balancing Equations Packet and Bestseller Lists
5. Accessing Balancing Equations Packet Free and Paid eBooks
 - Balancing Equations Packet Public Domain eBooks

- Balancing Equations Packet eBook Subscription Services
- Balancing Equations Packet Budget-Friendly Options
- 6. Navigating Balancing Equations Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Balancing Equations Packet Compatibility with Devices
 - Balancing Equations Packet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Balancing Equations Packet
 - Highlighting and Note-Taking Balancing Equations Packet
 - Interactive Elements Balancing Equations Packet
- 8. Staying Engaged with Balancing Equations Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Balancing Equations Packet
- 9. Balancing eBooks and Physical Books Balancing Equations Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Balancing Equations Packet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Balancing Equations Packet
 - Setting Reading Goals Balancing Equations Packet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Balancing Equations Packet
 - Fact-Checking eBook Content of Balancing Equations Packet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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