

Balancing Act

Name _____

Atoms are not _____ or _____ during a chemical reaction. Scientists know that there must be the _____ number of atoms on each _____ of the _____. To balance the chemical equation, you must add _____ in front of the chemical formulas in the equation. You cannot _____ or _____ subscripts!

1) Determine number of atoms for each element.

2) Pick an element that is not equal on both sides of the equation.

3) Add a coefficient in front of the formula with that element and adjust your counts.

4) Continue adding coefficients to get the same number of atoms of each element on each side.



Mg =

Mg =

O =

O =

Try these:



Ca =

Ca =

O =

O =



N =

N =

H =

H =



Cu =

Cu =

O =

O =

C =

C =



H =

H =

O =

O =

Balancing Act Chemical Equations Worksheet Answers

Chris McMullen



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Te HS&T a Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 *A Balancing Act! How to Balance a Chemical Equation and the Law of Conservation of Mass | Grade 6-8 Physical Science* Baby Professor, 2024-04-15 Explore the intriguing principles of chemical balance with this essential guide perfect for middle school educators homeschooling parents and librarians This book unravels the Law of Conservation of Mass and demystifies the process of balancing chemical equations underscoring their significance in the US STEM curriculum Gain insights into the immutable nature of mass in chemical reactions and master the art of equation balancing to foster a deep understanding of chemical transformations a must have resource for promoting scientific literacy in young learners [Cooperative Learning in the Chemistry Classroom](#)

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