Balancing Equations

Constituted Six more

Section It

Balance the following chemical equations below. [8]

(a)
$$1 C_1 H_0 O_2 (1) + 5 O_2 (g) \rightarrow 4 H_2 O (g) + 4 CO_2 (g)$$

(b)
$$1 \text{ Al}_2(CO_3)_1(s) \rightarrow 1 \text{ Al}_2O_3(s) + 3 \text{ CO}_2(g)$$

(c)
$$2 \text{ HNO}_2(aq) + 1 \text{ Ni}(s) \rightarrow 1 \text{ Ni}(NO_2)_2(aq) + 1 \text{ Hz}(q)$$

(d)
$$\underline{1}$$
Mn (s) + $\underline{2}$ CuCl (eq) \rightarrow $\underline{2}$ Cu (s) + $\underline{1}$ MnCl₂ (s)

Section 2:

Construct balanced chemical equations for the reactions below. [4]

 (a) Zinc undergoes displacement reaction with aqueous silver nitrate to produce zinc nitrate and silver

 (b) Calcium Carbonate decomposes to give calcium oxide and carbon dioxide

Balancing Equations 2 Chemistry Answer Key

Richard F. Brinckeroff

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