

Balancing Equations Worksheet

- 1) ____ Na_3PO_4 + ____ KOH \rightarrow ____ NaOH + ____ K_3PO_4
- 2) ____ MgF_2 + ____ Li_2CO_3 \rightarrow ____ MgCO_3 + ____ LiF
- 3) ____ P_4 + ____ O_2 \rightarrow ____ P_2O_3
- 4) ____ RbNO_3 + ____ BeF_2 \rightarrow ____ $\text{Be}(\text{NO}_3)_2$ + ____ RbF
- 5) ____ AgNO_3 + ____ Cu \rightarrow ____ $\text{Cu}(\text{NO}_3)_2$ + ____ Ag
- 6) ____ CF_4 + ____ Br_2 \rightarrow ____ CBr_4 + ____ F_2
- 7) ____ HCN + ____ CuSO_4 \rightarrow ____ H_2SO_4 + ____ $\text{Cu}(\text{CN})_2$
- 8) ____ GaF_3 + ____ Cs \rightarrow ____ CsF + ____ Ga
- 9) ____ BaS + ____ PtF_2 \rightarrow ____ BaF_2 + ____ PtS
- 10) ____ N_2 + ____ H_2 \rightarrow ____ NH_3
- 11) ____ NaF + ____ Br_2 \rightarrow ____ NaBr + ____ F_2
- 12) ____ $\text{Pb}(\text{OH})_2$ + ____ HCl \rightarrow ____ H_2O + ____ PbCl_2
- 13) ____ AlBr_3 + ____ K_2SO_4 \rightarrow ____ KBr + ____ $\text{Al}_2(\text{SO}_4)_3$
- 14) ____ CH_4 + ____ O_2 \rightarrow ____ CO_2 + ____ H_2O
- 15) ____ Na_3PO_4 + ____ CaCl_2 \rightarrow ____ NaCl + ____ $\text{Ca}_3(\text{PO}_4)_2$
- 16) ____ K + ____ Cl_2 \rightarrow ____ KCl
- 17) ____ Al + ____ HCl \rightarrow ____ H_2 + ____ AlCl_3
- 18) ____ N_2 + ____ F_2 \rightarrow ____ NF_3
- 19) ____ SO_2 + ____ Li_2Se \rightarrow ____ SSe_2 + ____ Li_2O
- 20) ____ NH_3 + ____ H_2SO_4 \rightarrow ____ $(\text{NH}_4)_2\text{SO}_4$

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A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Speedy Publishing' text.

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A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05
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authority patrol or any other institution to an individual stating that the secondary does not take any criminal proceedings
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