

Balancing Equations Worksheet

- 1) ____ Na_3PO_4 + ____ KOH \rightarrow ____ NaOH + ____ K_3PO_4
- 2) ____ MgF_2 + ____ Li_2CO_3 \rightarrow ____ MgCO_3 + ____ LiF
- 3) ____ P_4 + ____ O_2 \rightarrow ____ P_2O_3
- 4) ____ RbNO_3 + ____ BeF_2 \rightarrow ____ $\text{Be}(\text{NO}_3)_2$ + ____ RbF
- 5) ____ AgNO_3 + ____ Cu \rightarrow ____ $\text{Cu}(\text{NO}_3)_2$ + ____ Ag
- 6) ____ CF_4 + ____ Br_2 \rightarrow ____ CBr_4 + ____ F_2
- 7) ____ HCN + ____ CuSO_4 \rightarrow ____ H_2SO_4 + ____ $\text{Cu}(\text{CN})_2$
- 8) ____ GaF_3 + ____ Cs \rightarrow ____ CsF + ____ Ga
- 9) ____ BaS + ____ PtF_2 \rightarrow ____ BaF_2 + ____ PtS
- 10) ____ N_2 + ____ H_2 \rightarrow ____ NH_3
- 11) ____ NaF + ____ Br_2 \rightarrow ____ NaBr + ____ F_2
- 12) ____ $\text{Pb}(\text{OH})_2$ + ____ HCl \rightarrow ____ H_2O + ____ PbCl_2
- 13) ____ AlBr_3 + ____ K_2SO_4 \rightarrow ____ KBr + ____ $\text{Al}_2(\text{SO}_4)_3$
- 14) ____ CH_4 + ____ O_2 \rightarrow ____ CO_2 + ____ H_2O
- 15) ____ Na_3PO_4 + ____ CaCl_2 \rightarrow ____ NaCl + ____ $\text{Ca}_3(\text{PO}_4)_2$
- 16) ____ K + ____ Cl_2 \rightarrow ____ KCl
- 17) ____ Al + ____ HCl \rightarrow ____ H_2 + ____ AlCl_3
- 18) ____ N_2 + ____ F_2 \rightarrow ____ NF_3
- 19) ____ SO_2 + ____ Li_2Se \rightarrow ____ SSe_2 + ____ Li_2O
- 20) ____ NH_3 + ____ H_2SO_4 \rightarrow ____ $(\text{NH}_4)_2\text{SO}_4$

Balancing Equations Practice Key Ch 8

Peter P. Berlow



Balancing Equations Practice Key Ch 8:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields *Foundations of College Chemistry* Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis *The Nurse, The Math, The Meds - E-Book* Joyce L. Mulholland, Susan Turner, 2014-10-01 Use the simplicity of the dimensional analysis method to minimize drug calculation errors The Nurse The Math The Meds 3rd Edition helps you overcome any

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