# Single Replacement Reactions

27. \_\_AgNO<sub>3</sub> + \_\_Ni 
$$\rightarrow$$
 \_\_Ni(NO<sub>3</sub>)<sub>2</sub> + \_\_Ag  
28. \_\_AlBr<sub>3</sub> + \_\_Cl<sub>2</sub>  $\rightarrow$  \_\_AlCl<sub>3</sub> + \_\_Br<sub>2</sub>  
29. \_\_Nal + \_\_Br<sub>2</sub>  $\rightarrow$  \_\_NaBr + \_\_l<sub>2</sub>  
30. \_\_Ca + \_\_HCl  $\rightarrow$  \_\_CaCl<sub>2</sub> + \_\_H<sub>2</sub>  
31. \_\_Mg + \_\_HNO<sub>3</sub>  $\rightarrow$  \_\_Mg(NO<sub>3</sub>)<sub>2</sub> + \_\_H<sub>2</sub>  
32. \_\_Zn + \_\_H<sub>2</sub>SO<sub>4</sub>  $\rightarrow$  \_\_ZnSO<sub>4</sub> + \_\_H<sub>2</sub>  
33. \_\_K + \_\_H<sub>2</sub>O  $\rightarrow$  \_\_KOH + \_\_H<sub>2</sub>  
34. \_\_Na + \_\_H<sub>2</sub>O  $\rightarrow$  \_\_NaOH + \_\_H<sub>2</sub>

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