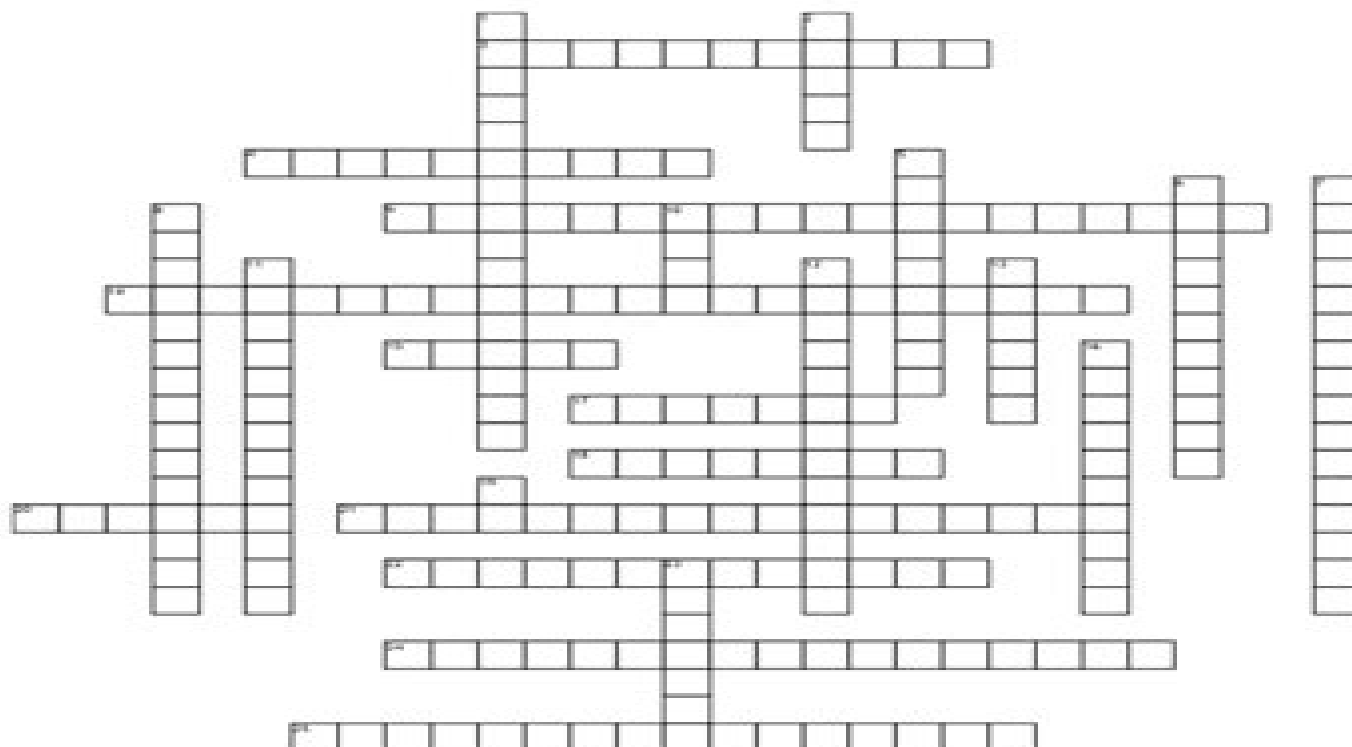


Solutions, Acids, and Bases



Across

3. Ammonia, antacids, egg whites, and baking powder are examples of _____.
4. Maximum amount of a solute that can be dissolved in a given amount of solvent.
9. Any solution that can dissolve more solute at a given temperature.
14. Solution that contains too much solute.
15. Produce hydrogen ions in water.
17. Substance doing the dissolving.
18. Physically mixed-sitting next to each other, but not joined chemically.
20. Substance that is dissolved.

21. Large amount of solute in a solution.

22. Increased concentrations of solute in a solution will lower the _____.

24. Solution that contains all the solute it can hold at a given temperature.

25. The time it takes a solute to dissolve in a solvent in a given solution.

Down

1. Solutions are mad of H^+ ion and an anion.
2. Produce hydroxide ions in water.
3. Unable to be dissolved.
6. Citric acid in citrus fruits.

7. Water is called this because it dissolves more substances than others.

8. A curve on a graph that shows how much solute can be dissolved.

10. The time it takes an action to occur.

11. A mixture that does not have a uniform composition.

12. The amount of solute in a given amount of solvent.

13. Low amount of solute in solution.

16. A mixture that has a uniform composition.

19. Measurement of amount of hydrogen ions.

23. The number 7 on the pH scale.

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Stuart Ira Fox, Laurence G. Thouin



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