

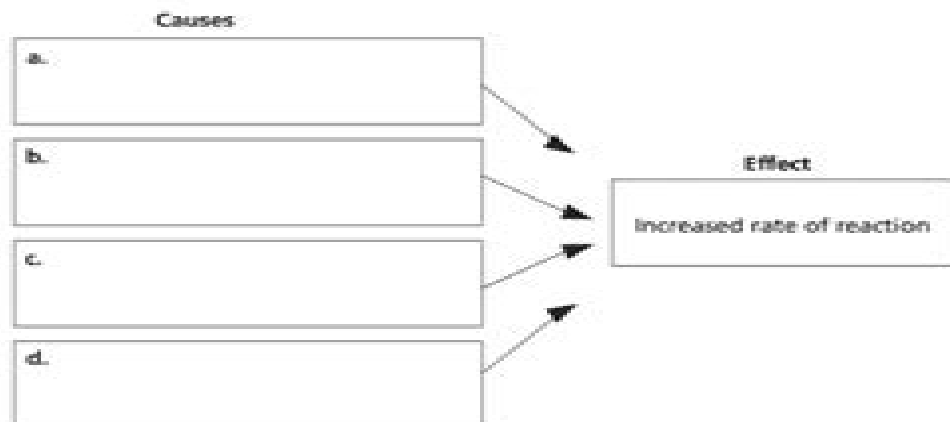
Chemical Reactions • Guided Reading and Study

Controlling Chemical Reactions

This section explains how energy is related to chemical reactions. It also describes how the rates of chemical reactions can be controlled.

Use Target Reading Skills

As you read, identify the factors that can cause the rate of a chemical reaction to increase. Write the information in the graphic organizer below.



Energy and Reactions

1. The _____ is the minimum amount of energy needed to start a chemical reaction.
2. Is the following sentence true or false? All chemical reactions need a certain amount of activation energy to get started.

3. In a reaction that makes water from hydrogen gas and oxygen gas, where does the activation energy come from?

4. A reaction that releases energy is called a(n) _____.
5. A reaction that absorbs energy is called a (n) _____.

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Prentice Hall Science Explorer Michael J. Padilla, 2002

Prentice Hall Physical Science Concepts in Action

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