

ELECTRICAL CIRCUITS AND SIMULATION LAB

**DEPARTMENT OF
ELECTRICAL AND ELECTRONICS ENGINEERING**

ACADEMIC YEAR 2011-2012

II B.Tech II SEMESTER

Diploma Electrical Circuits Simulation Lab Manual

Victor M. Corman



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web the balanced chemical equation for this reaction is $\text{mg s} + 2\text{hno}_3 \text{ aq} \rightarrow \text{mg no}_3 2 \text{ aq} + \text{h}_2 \text{ g}$ in this equation the s symbolizes
solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the
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 $\text{Mg}^{2+} + 2 \text{OH}^- + 2 \text{H}^+ + 2 \text{NO}_3^- \rightarrow \text{Mg}^{2+} + 2 \text{NO}_3^- + 2 \text{H}_2\text{O}$ net ionic equation $2 \text{OH}^- + 2 \text{H}^+ \rightarrow 2 \text{H}_2\text{O}$ sulfuric acid and lithium hydroxide
balanced total equation $\text{H}_2\text{SO}_4 \text{aq} + 2 \text{LiOH aq} \rightarrow \text{Li}_2\text{SO}_4 \text{aq} + 2 \text{H}_2\text{O}$