

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

1
2
Hydrogen
Helium
Group 1

Atomic Mass	1.1
Atomic Number	1.2
Electron Configuration	1.3
Chemical Name	1.4
Chemical Symbol	1.5

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1.1. It appears in the periodic table.
- _____ 1.2. It has no line periodic table.
- _____ 1.3. Group 1 elements.
- _____ 1.4. Elements that are shiny and conduct electricity.
- _____ 1.5. Group 1 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 1.6. There are two representative elements of elements.
- _____ 1.7. Most elements have lines in the periodic table for metals.
- _____ 1.8. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 1.9. Group 2 elements are the alkaline earth metals.
- _____ 1.10. Group 1 elements are highly reactive elements known as alkalis.
- _____ 1.11. Group 1 elements are very reactive elements known as reactive metals.
- _____ 1.12. Alkaloids have properties of both metals and non-metals.

Chapter Study Guide For Content Mastery

Sheila Alber Morgan



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