

Specific Heat

Specific Heat

Change in Energy

If $w=0$, then $c = \frac{1}{m} \frac{Q}{\Delta t}$

$$c = \frac{1}{m} \frac{\Delta E}{\Delta T}$$

Specific heat is a material property

mass

Change in Temperature

In words: The **specific heat** is equal to one over **mass** multiplied by the **change in energy** divided by the **change in temperature**

Chapter Thermal Energy And Heat Calculation With Specific

Augustin Privat-Deschanel



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Energy, Matter, and Change William B. Tucker, 2024-12-27 This textbook serves as an introduction to the field of chemistry aimed at secondary school students and it assumes no prior knowledge on the readers part As an introductory text the book emphasizes fundamental skills that are necessary for chemistry and science generally This includes an emphasis on good writing and a focus on problem solving with problems incorporated throughout the text To help prepare students to pursue chemistry further all information presented is in accord with the International Union of Pure and Applied Chemistry's style and technical guidelines and supported through citations to the primary literature The Open Access version of this book available at <http://www.taylorfrancis.com> has been made available under a Creative Commons Attribution Non Commercial No Derivatives CC BY NC ND 4.0 license

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