

13. HYDROCARBONS

Organic compounds containing carbon and hydrogen atoms only are called hydrocarbons. Depending on the types of C-C bond, they can be classified into three – saturated, unsaturated and aromatic hydrocarbons. Saturated hydrocarbons are also called alkanes. They contain only C-C single bonds. Unsaturated hydrocarbons contain atleast one carbon-carbon double bond (alkene) or carbon-carbon triple bond (alkyne). Aromatic hydrocarbons are a special type of cyclic compounds. They are also called arenes.

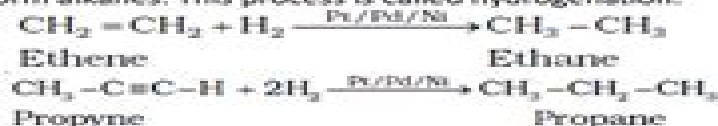
ALKANES

Alkanes are saturated open chain hydrocarbons containing carbon-carbon single bonds. They form a homologous series. Their general molecular formula is C_nH_{2n+2} . In alkanes, all the C atoms are sp^3 hybridised. So each C atom has a regular tetrahedral shape.

Alkanes do not react with acids, bases and other reagents under normal conditions. So they are also called paraffins. (In Latin paraffin means little affinity).

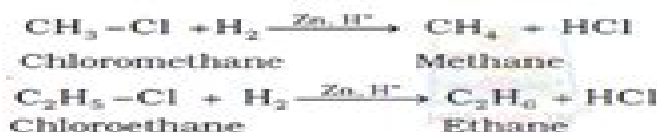
Preparation of alkanes

1. **From unsaturated hydrocarbons:** Alkenes and alkynes add Hydrogen in presence of finely divided catalysts like Ni, Pd or Pt to form alkanes. This process is called hydrogenation.



2. **From alkyl halides:**

- a) Alkyl halides on reduction with zinc and dil. HCl, we get alkanes.



- b) **Wurtz reaction:**

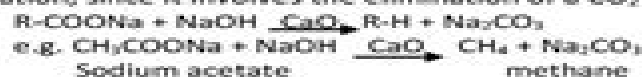
Alkyl halides react with metallic sodium in dry ether to form alkanes. This reaction is known as Wurtz reaction. The alkane formed contains double the number of C atoms than that present in the alkyl halide. So this method is used for the preparation of alkanes with even number of carbon atoms.



When two different alkyl halides are used, we get a mixture of alkanes.

3. **From carboxylic acids:**

a) **Decarboxylation:** Sodium salt of carboxylic acids (R-COONa) on heating with soda lime (a mixture of NaOH and CaO), we get an alkane containing one carbon atom less than that of the carboxylic acid. This process is known as decarboxylation, since it involves the elimination of a CO_2 molecule from the carboxylic acid.



b) **Kolbe's Electrolytic method:** An aqueous solution of sodium or potassium salt of a carboxylic acid on electrolysis gives alkane containing even number of carbon atoms.

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web jul 29 2023 express your gratitude towards the guests for attending by using a phrase such as it s so wonderful to see

you all here on this sunny day 2 for an event with close friends and family more informal language might be appropriate include a few jokes and keep your speech more lighthearted

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web in the case of a welcoming ceremony the welcome speech is considered to be one of the most important parts of the whole gathering in fact it is the most crucial aspect of the entire event this is because the welcome speech serves several purposes first it introduces the guests to the host

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web a welcome speech as the name suggests is a type of speech that is mostly given by the host or hosts at the beginning of an event or a ceremony this is where the speaker is showing some gratitude and appreciation for the people who

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web depending on the event and the type of acknowledgment the content of the speech may vary from showing appreciation for the award simply saying i m happy to hold this award in my hands to giving thanks to all who have contributed to gaining a particular award

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web jan 17 2023 notable examples of award recipients 35 in this environment it is important for us as the home team to be nimble savvy and bold in engaging our singaporeans and in how we work 36 so let me share some examples from among our award recipients 37

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web welcome speech for award ceremony an award ceremony is an event at which someone is given an award for their achievements this could be for their work in a particular field their sports performance or any other area where they have excelled

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welcome speech for award ceremony for students and a - Oct 23 2023

web apr 4 2023 welcome everyone it is my pleasure to welcome the honorable judges parents and our dear students i feel honored to share the achievements of the brilliant students of our school this ceremony will award the meritorious students and it also marks the progress and success of our school