

CHAPTER 6

STUDY GUIDE

Ionic Compounds and Metals

Lesson 6.1 Ion Formation

In your textbook, read about chemical bonds and formation of ions.

Use each of the terms below just once to complete the passage.

| | | | | |
|---------------|-----------|-----------------------------|---------|-------------|
| chemical bond | electrons | energy level | ions | noble gases |
| nucleus | octet | pseudo-noble gas formations | valence | |

The force that holds two atoms together is called a(n) (1) Chemical Bond. Such an attachment may form by the attraction of the positively charged (2) Nucleus of one atom for the negatively charged (3) Electron of another atom, or by the attraction of charged atoms, which are called (4) Ions. The attractions may also involve (5) Valence electrons, which are the electrons in the outermost (6) Energy level. The (7) Noble Gas are a family of elements that have very little tendency to react. Most of these elements have a set of eight outermost electrons, which is called a stable (8) Octet. The relatively stable electron structures developed by loss of electrons in certain elements of groups 3, 4, 13, and 14 are called (9) Pseudo-Noble gas formations.

For each statement below, write *true* or *false*.

- | | |
|--------------|--|
| <u>False</u> | 10. A positively charged ion is called an anion. |
| <u>True</u> | 11. Elements in group 1 lose their one valence electron, forming an ion with a 1+ charge. |
| <u>False</u> | 12. Elements tend to react so that they acquire the electron structure of a halogen. |
| <u>True</u> | 13. A sodium atom tends to lose one electron when it reacts. |
| <u>True</u> | 14. The electron structure of a zinc ion (Zn^{2+}) is an example of a pseudo-noble gas formation. |
| <u>False</u> | 15. A Cl^- ion is an example of a cation. |
| <u>True</u> | 16. The ending <i>-ide</i> is used to designate an anion. |
| <u>False</u> | 17. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions. |

Chapter Study Guide Ionic Compounds And Metals

WJ Hussar



Chapter Study Guide Ionic Compounds And Metals:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
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Table of Contents Chapter Study Guide Ionic Compounds And Metals

1. Understanding the eBook Chapter Study Guide Ionic Compounds And Metals
 - The Rise of Digital Reading Chapter Study Guide Ionic Compounds And Metals
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Study Guide Ionic Compounds And Metals
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter Study Guide Ionic Compounds And Metals
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Study Guide Ionic Compounds And Metals

- Personalized Recommendations
 - Chapter Study Guide Ionic Compounds And Metals User Reviews and Ratings
 - Chapter Study Guide Ionic Compounds And Metals and Bestseller Lists
5. Accessing Chapter Study Guide Ionic Compounds And Metals Free and Paid eBooks
- Chapter Study Guide Ionic Compounds And Metals Public Domain eBooks
 - Chapter Study Guide Ionic Compounds And Metals eBook Subscription Services
 - Chapter Study Guide Ionic Compounds And Metals Budget-Friendly Options
6. Navigating Chapter Study Guide Ionic Compounds And Metals eBook Formats
- ePub, PDF, MOBI, and More
 - Chapter Study Guide Ionic Compounds And Metals Compatibility with Devices
 - Chapter Study Guide Ionic Compounds And Metals Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chapter Study Guide Ionic Compounds And Metals
 - Highlighting and Note-Taking Chapter Study Guide Ionic Compounds And Metals
 - Interactive Elements Chapter Study Guide Ionic Compounds And Metals
8. Staying Engaged with Chapter Study Guide Ionic Compounds And Metals
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter Study Guide Ionic Compounds And Metals
9. Balancing eBooks and Physical Books Chapter Study Guide Ionic Compounds And Metals
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter Study Guide Ionic Compounds And Metals
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter Study Guide Ionic Compounds And Metals
- Setting Reading Goals Chapter Study Guide Ionic Compounds And Metals
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter Study Guide Ionic Compounds And Metals

- Fact-Checking eBook Content of Chapter Study Guide Ionic Compounds And Metals
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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