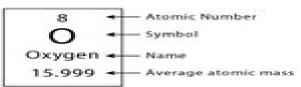
Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- a) What type of charge does a proton have? ______
- b) What type of charge does a neutron have?
- What type of charge does an electron have?
- Which two subatomic particles are located in the nucleus of an atom?
- 2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.



6	#P _
Carbon	#N
12.011	#E.





19	#P
K Potassium	#N
39.098	#E

3. Complete the table using your knowledge of the periodic table.

Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na			11		12
к		39		19	
			38		50
F				9	10
	20	40		20	
	50			50	69
T.	53	127		ĺ i	

Chapter Test Atom And Periodic Table

Sharon J. Sherman

Chapter Test Atom And Periodic Table:

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