

## Thermochemistry Review Part 1

**Definition Questions:** you should be able to use or apply the following terms. Be able to write complete definitions for,

energy	temperature	standard state
potential energy	heat	standard heat of formation
kinetic energy	specific heat capacity	exothermic
enthalpy	endothermic	Law of conservation of Energy
thermal	Hess's Law	isolated, closed, open system

1. The specific heat capacity of diamond is  $0.5050 \text{ J/g}^\circ\text{C}$ . How much energy is required to heat  $25.0 \text{ g}$  of diamond from  $10.5^\circ\text{C}$  to  $15.6^\circ\text{C}$ ?

$$q = m \times c \times \Delta t$$

$$= 25 \text{ g} \times 0.5050 \text{ J/g}^\circ\text{C} \times (15.6 - 10.5^\circ\text{C}) = 64.39 \text{ J}$$

2. A piece of aluminum with a mass of  $50 \text{ g}$  and an initial temperature of  $90^\circ\text{C}$  is placed into  $100 \text{ mL}$  of water at a temperature of  $25^\circ\text{C}$ . The temperature of water rises to  $31.3^\circ\text{C}$ . Determine the specific heat capacity of aluminum.

$Q \text{ lost (Al)} = Q \text{ gained (Water)}$

$$-(mc\Delta t) = +mc \Delta t$$

$$-(50 \times c \times (31.3 - 90)) = (100 \times 4.18 \times (31.3 - 25))$$

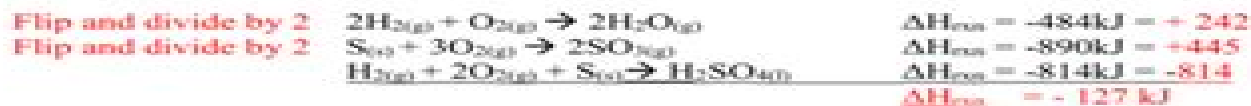
$$-(-2935 c) = 2633.4$$

$$c = 2633.4 / 2935 = 0.897 \text{ J/g}^\circ\text{C}$$

3. How much energy is absorbed by  $300 \text{ g}$  of methanol,  $\text{CH}_3\text{OH}$ , as it evaporates? Given: The molar heat of vaporization is  $35.3 \text{ kJ/mol}$ .

$q = ?$	$\Delta H = q/n$	$n \text{ CH}_3\text{OH} = m/\text{MM}$
$m = 300 \text{ g}$	$q = H \times n$	$n = 300 \text{ g} / 32 \text{ g/mol}$
$\Delta H = 35.3 \text{ kJ/mol}$	$q = (35.3 \text{ kJ} \times 9.375 \text{ mol})$	$n = 9.375 \text{ mol}$
	$q = 332.81 \text{ kJ}$	

4. What is the molar enthalpy of the formation of  $1 \text{ mol H}_2\text{SO}_{4(l)}$  given the following information?



# Chapter Thermochemistry Review Answers

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### 5.5 - Hypothesis Testing for Two-Sample Proportions

We are now going to develop the hypothesis test for the difference of two proportions for independent samples. The hypothesis test follows the same steps as ...

#### Two-Sample t-Test | Introduction to Statistics

The two-sample t-test is a method used to test whether the unknown population means of two groups are equal or not. Learn more by following along with our ...

#### 1.3.5.3. Two-Sample t-Test for Equal Means Purpose: Test if two population means are equal, The two-sample t-test (Snedecor and Cochran, 1989) is used to determine if two population means are equal.

#### 2 Sample t-Test (1 tailed) Suppose we have two samples of ceramic sherd thickness collected from an archaeological site, where the two samples are easily distinguishable by the use of.

#### Two sample t-test: SAS instruction Note that the test is two-sided (sides=2), the significance level is 0.05, and the test is to compare the difference between two means ( $\mu_1 - \mu_2$ ) against 0 ( $H_0$ ...

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